

Development and Validation Method for the Determination of Sildenafil Citrate Tablets by using UV-Spectrophotometer in Pharmaceutical Formulation

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Abstract

The objective of this paper is to describe the optimization, validation, and application of spectrophotometric technique for determination of Sildenafil Citrate in their pharmaceutical formulation (tablets). In this paper, a simple, rapid, accurate and sensitive spectrophotometric method has been developed and validated. The method is a direct spectrophotometric method based on thawed of sildenafil citrate in diluted hydrochloric acid. The maximum absorption wavelength for the determination of Sildenafil citrate was found to be 295 nm. Under the optimized condition, Beer's law was obeyed in the concentration range from 5.0-40.0 µg/ml.

Keywords: Sildenafil citrate; Absorption; UV-spectrophotometer

Introduction

Sildenafil (SC) is a compound of the Pyrazolo-pyrimidinyl-methylpiperazine class. It is 5-[2-ethoxy-5-(4-methyl piperazin-1-yl sulphonyl) phenyl] -1-Methyl-3-propyl-1, 6-dihydro-7Hpyrazolo [4,3-d] pyrimidin-7-one, having empirical formula C₂₂H₃₀N₆O₄S and molecular Weight 661.71 [1]. Sildenafil Citrate is a white to off-white crystalline powder [2]. It is used not only for erectile dysfunction treatment but also for pulmonary arterial hypertension [3].

In healthy volunteers, sildenafil is extensively absorbed; however, rapid first-pass metabolism limits the absolute bioavailability to approximately 40%. C_{max} is reached within 30-120 minutes of oral administration in the fasted state, and the terminal half-life is about 4 hours [4].

The most common side effects of Sildenafil citrate are a headache, facial flushing, and upset stomach. [5].

Its structural formula is given in FIG. 1.

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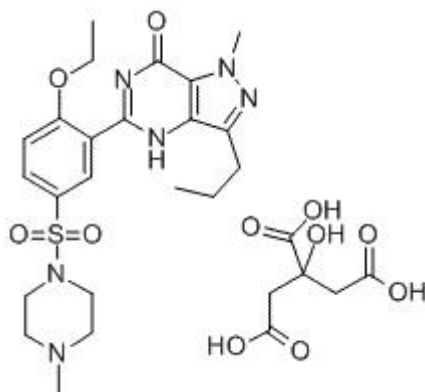


FIG. 1. Structure of sildenafil citrate.

Sildenafil citrate, a specific phosphodiesterase-5 inhibitor [6], is increasingly used for pulmonary hypertension in pregnancy [7]. Sildenafil Citrate is also emerging as a potential candidate for the treatment of intra-uterine growth retardation and for premature labor [8]. Its effects in the fetoplacental circulation are not known [9]. Our objectives were to determine whether phosphodiesterase-5 is present in the human fetoplacental circulation and to characterize the effects and mechanisms of action of Sildenafil Citrate in this circulation [10,11].

Validation is defined as finding or testing the truth of something. The objective of validation of an analytical procedure is to demonstrate that it is suitable for its intended purpose [12].

Analytical methods need to be validated, verified, or revalidated in the following instances:

- Before initial use in routine testing
- When transferred to another laboratory
- Whenever the conditions or method parameters for which the method has been validated change (for example, an instrument with different characteristics or samples with a different matrix) and the change is outside the original scope of the method [12]

Materials and Methods

Collection of the sample

All pharmaceutical formulation and samples were taken from local markets.

Reagents and solutions

All chemicals used are Analytical Grade (AG), and spectroscopic organic solvents. Deionized water was used for preparing solutions. Sildenafil citrate working standard, (purity 99.57%) was purchased from Rakshit Drugs Private Limited. Hydrochloric Acid (SDFCL, India), sp.gr:1.18 g/ml, percentage: 37.5% Hydrochloric acid 0.1 M.

Instrumentation

The following instruments, equipment or apparatus were used during the course of this work.

UV-Visible Spectrophotometer model UV-1800 (SHIMADZU, KYOTO, JAPAN) equipped with 1 cm quartz cell, was used in the quantitative analysis of assay, the absorption spectra of test and reference solution were recorded in 1 cm quartz cells at

many λ max. FTIR (Fourier Transform Infra-Red spectrophotometer), (SHIMADZU, KYOTO, JAPAN), Model FTIR-8400s. Electronic Sensitive balance (SHIMADZU, KYOTO, JAPAN), Type A \times 120, Capacity 120 g, Readability 0.1 mg.

Preparation of 0.1M hydrochloric acid solution

8.0 ml of concentrated hydrochloric acid was taken accurately and transferred to 1000 ml volumetric flask containing about 300 ml of deionized water, mixed well, cooled, and completed the volume to the mark with the same diluent and mixed.

Solubility studies of Sildenafil Citrate for UV-Spectrophotometer analysis: Solubility of SIL was determined at $(28 \pm 2)^\circ\text{C}$. An excess amount of the drug was taken into 25 ml volumetric flasks each containing different concentrations of diluted hydrochloric acid such as (0.01, 0.1, 0.5, 1.0) M, suitable solvent for sildenafil citrate (SIL) chosen (0.1 M HCl) according to solubility of drug and high intensity (absorbance) of the drug.

Determination of maximum absorption

Determination of the wavelength of maximum absorption (λ max) of sildenafil citrate: 10 mg of Sildenafil citrate working standard was accurately weighed and transferred to a 100 ml volumetric flask, 5.0 ml of 0.1N HCl was added to the weight and sonicated for 2 minutes for dissolution. The solution was kept aside to cool and completed to the mark with the same solvent to give a 100 $\mu\text{g/ml}$ stock solution which was diluted suitably to produce (0.6, 1.0, 10.0, 25.0 and 50.0) $\mu\text{g/ml}$ of SILC. This solution was scanned in the spectrum mode from 200-800 nm. From the spectrum of the drug obtained λ max of SIL was determined at 295 nm.

Determination of the wavelength of maximum absorption (λ max) of placebo of sildenafil citrate: From the total weight of tablet placebo PLC composition 274.83 mg accurately weighed 10.0 mg of PLC Placebo of Sildenafil Citrate tablets compositions and transferred to 100 ml volumetric flask, minimum amount (about 5.0 ml) of 0.1N HCl was added to the weight of PLC and dissolved by sonication for 2.0 minute for dissolution, cooled in room temperature and completed to the mark with the same solvent to give 100 $\mu\text{g/ml}$ stock solution which was diluted suitably to produce (5.0, 10.0, 25.0 and 50) $\mu\text{g/ml}$ of PLC. This solution was scanned in the spectrum mode from 200-800 nm. From the spectrum of the PLC composition, there is no spectrum obtained in the λ max of Sildenafil Citrate drug 271 nm (FIG. 2).

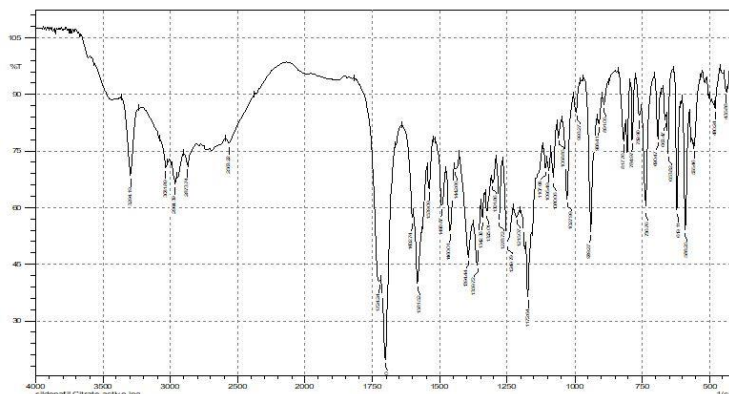


FIG. 2. FTIR spectra of sildenafil citrate STD (reference).

Determination of the wavelength of maximum absorption (λ_{\max}) of 0.1N HCl (Sildenafil Citrate solvent): 3.5 ml of 0.1N HCl scanned in the UV-Spectrophotometer in the range of 200-800 nm, the maximum absorption wavelength of the diluted acid was found to be 208 nm (FIG. 3).

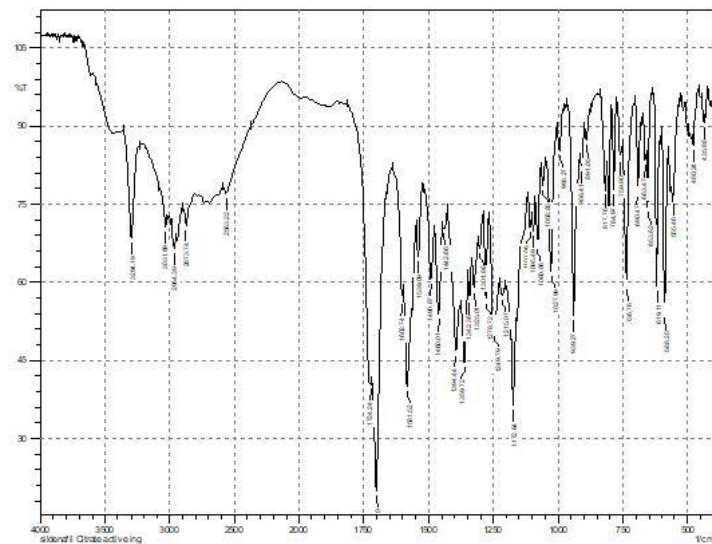


FIG. 3. Identification of sildenafil citrate and excipient (Tablets 25 mg).

Preparation of standard stock solution of Sildenafil Citrate for UV-spectrophotometer analysis

50 mg of sildenafil citrate working standard (99.57% purity) was accurately weighed and dissolved in 0.1M HCl, sonicated for 60 seconds for dissolution, transferred quantitatively into 50 ml volumetric flask, completed the volume to the mark with the same solvent and mixed well, 1.0 ml of this solution was transferred to 100 ml volumetric flask, completed to the mark with 0.1M HCl to get (15,20,25,30,35,40,105) $\mu\text{g/ml}$ and absorbance was measured at 295 nm.

Preparation of sample solution of sildenafil citrate for UV-spectrophotometer analysis

Equivalent to 50 mg of Sildenafil citrate from 20 tablets were finely powdered Sildenafil citrate tablets drugs and weighted, dissolved in 0.1M HCl, sonicated for 60 seconds for dissolution, then transferred quantitatively into 100 ml volumetric flask and completed the volume to the mark with the same solvent, mixed well and filtered through Whatman filter paper No. 41. 1 ml of this solution was transferred to 100 ml volumetric flask, diluted to the mark with 0.1M HCl to get 5.0 $\mu\text{g/ml}$ and measured the absorbance at 295 nm.

Results

Method development

The method is a direct spectrophotometric method based on thawed of sildenafil citrate in diluted hydrochloric acid.

Identification of sildenafil citrate STD

The below spectral of Sildenafil Citrate was found to exhibit characteristics absorption bands at 3240 cm^{-1} , 1627 cm^{-1} , 1620 cm^{-1} , 1180 cm^{-1} , 1100 cm^{-1} , 3600 cm^{-1} , 828 cm^{-1} , showing N-H, C=O, C=C, C-O, C-N, O-H, and aromatic substitution bands respectively for Sildenafil Citrate.

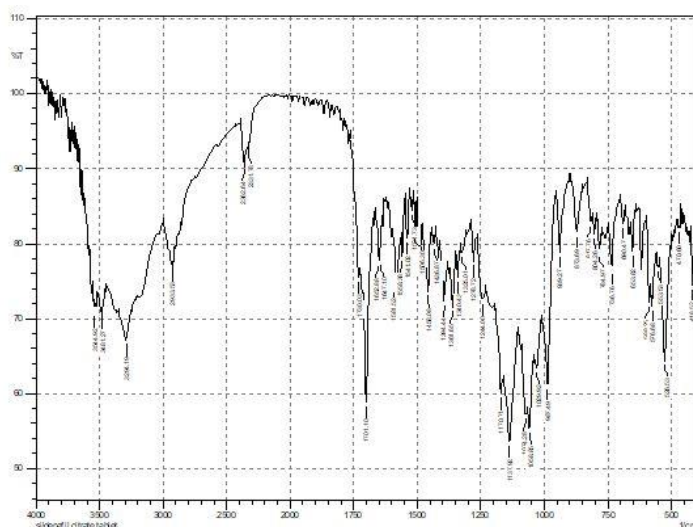


FIG. 4. FTIR spectra of sildenafil citrate tablets (sample).

The Infrared (IR) absorption spectrum of SIL tablets was generated using a Spectrum 100 Fourier transform-infrared attenuated total reflectance spectrometer (SHIMADZU 8400s). The spectrum, obtained at a wave-number range of 4000-418 cm^{-1} and a resolution of 4 cm^{-1} is described in FIG. 4. Band assignments for the resultant spectrum are summarized in TABLE 1 below [13].

TABLE 1. Infrared frequencies (cm^{-1}) and tentative assignments for sildenafil citrate complexes.

No.	Functional group(s)	Characteristic Absorption(s) range (cm^{-1})	Identified in this study (cm^{-1})
1	SO^2 stretch	1200-1100	1172.74
2	Aromatic C=C bond	1700-1500	1582.62
3	C=O stretch	1750-1680	1702.21
4	Saturated C-H stretch	2950-2850	2962.71
5	Unsaturated C-H stretch	3100-3010	3029.26
6	Secondary N-H stretch	3500-3300	3299.30
7	O-H stretch	3550-3200	3617.00

Method

For validation of analytical methods, the guidelines of the International Conference on the Harmonization (ICH) of Technical Requirements for the Registration of Pharmaceuticals for Human Use have recommended the accomplishment of accuracy tests, precision, specificity, the linearity of the method. ICH of Technical Requirements for the Registration of Pharmaceutical for Human Use (ICH) Q2B, 1996, validation of analytical procedures and methodology [12].

Determination of maximum absorption

Maximum absorption spectra of drug solvent (0.1N HCl): The absorption spectra of the SILC solvent (0.1 N HCl) which shown in FIG. 5 shows maximum absorbance of 0.1 N HCl at $\lambda=208$ nm, this maximum absorbance show there is no any interference between the excipients of the drug and active ingredient of the drug under study in the scanned spectral peaks region, while the absorption spectrum of diluted hydrochloric acid (0.1 N) shows no absorption. The peak at $\lambda=208$ nm was selected at maximum lambda for the solvent because it gives the highest absorption intensity, as indicated by the ϵ values (TABLE 2).

TABLE 2. The values of multi-wavelength absorbance of the drug Solvent (0.1 N HCl) shows many different absorbance values.

NO.	Wavelength (nm)	Absorbance
1	286.00	0.032
2	276.00	-0.102
3	262.00	-0.180
4	254.00	0.292
5	246.00	0.112
6	233.00	0.030
7	223.00	0.107
8	208.00	0.395
9	201.00	-0.477
10	280.00	-0.272
11	268.00	-0.493
12	260.00	-0.217
13	249.00	0.001
14	238.00	-0.085
15	229.00	-0.208
16	217.00	-0.112
17	214.00	-0.096
18	203.00	-0.803

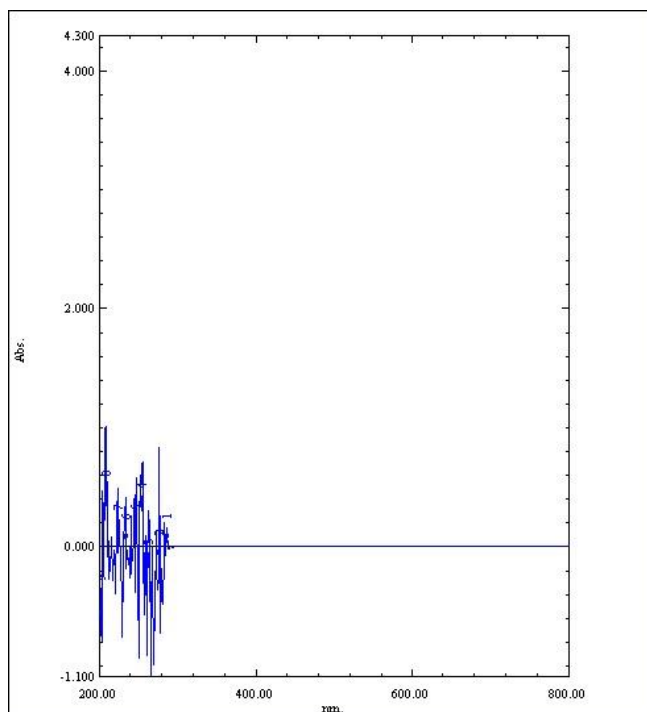


FIG. 5. Absorption spectra of sildenafil citrate tablets drug solvent 0.1 N HCl at room temperature showed in the scanned spectral region. The peak at $\lambda=208$ nm was selected as maximum absorbance of the solvent because it gives the highest absorption intensity.

Maximum absorption spectra of drug excipients (placebo)

Maximum absorption spectra of the SIL excipient (placebo) in FIG. 6 shows maximum absorbance $\lambda=271$ nm, this maximum absorbance show there is no any interference between the excipients of the drug and active ingredient of the drug under study in the scanned spectral peaks region, while the absorption spectrum of placebo shows no absorption. The peak at $\lambda=271$ nm was selected at maximum lambda for the placebo because it gives the highest absorption intensity, as indicated by the ϵ values (TABLE 3)

TABLE 3. Absorbance values of excipient (placebo) of SIL drug.

NO.	Wavelength (nm)	Absorbance
1	278.00	-0.080
2	271.00	0.096
3	255.00	-0.281
4	237.00	0.013
5	226.00	-0.010
6	218.00	-0.029
7	211.00	-0.032
8	201.00	-0.154
9	282.00	-0.108
10	276.00	-0.217
11	258.00	-0.353
12	252.00	-0.355
13	231.00	-0.137
14	214.00	-0.164
15	206.00	-0.331
16	203.00	-0.366

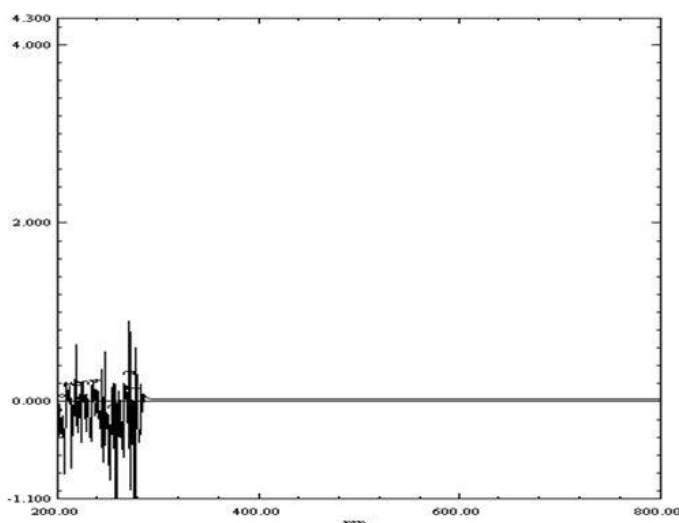


FIG. 6. Absorption spectra of placebo (excipients) of sildenafil citrate tablets at room temperature showed in the scanned spectral region. The peak at $\lambda=271$ nm was selected as maximum absorbance of the placebo because it gives the highest absorption intensity.

Maximum absorption spectra of drug standard solution

The absorption spectra of the SIL standard in 0.1 N HCl solvent (FIG. 7-9) show maximum absorbance at $\lambda=295$ nm, this maximum absorbance for different concentrations: (25, 33.3 and 50) $\mu\text{g/ml}$, while the absorption spectrum of SIL shows no absorption peaks in the scanned spectral region. The peak at $\lambda=295$ nm was selected because it gives the highest absorption intensity, as indicated by the ϵ values (TABLES 4-6).

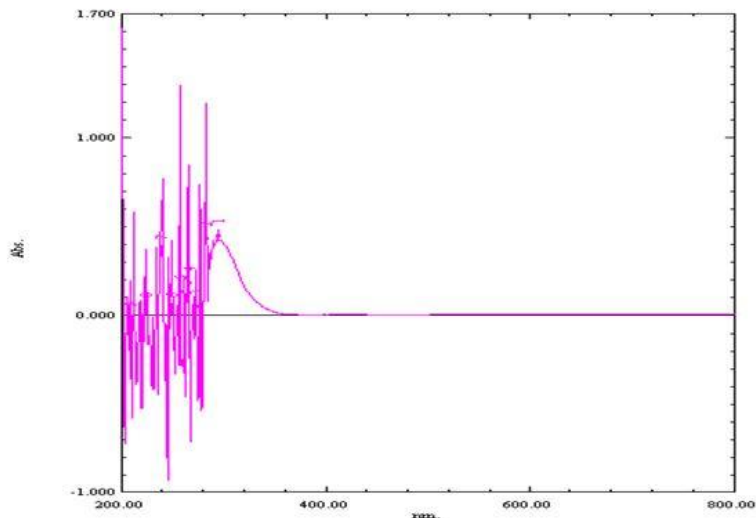


FIG. 7. Absorption spectra of 25 $\mu\text{g/ml}$ of sildenafil citrate pure drug at room temperature showed in the scanned spectral region. The peak at $\lambda=295$ nm was selected as maximum absorbance of the pure SIL because it gives the highest absorption intensity.

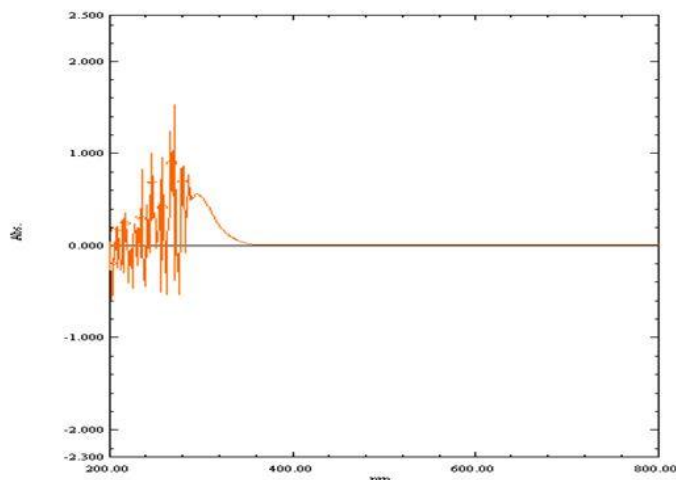


FIG. 8. Absorption spectra of 33.33 $\mu\text{g/ml}$ of sildenafil citrate pure drug at room temperature showed in the scanned spectral region. The peak at $\lambda=295$ nm was selected as maximum absorbance of the pure SIL because it gives the highest absorption intensity.

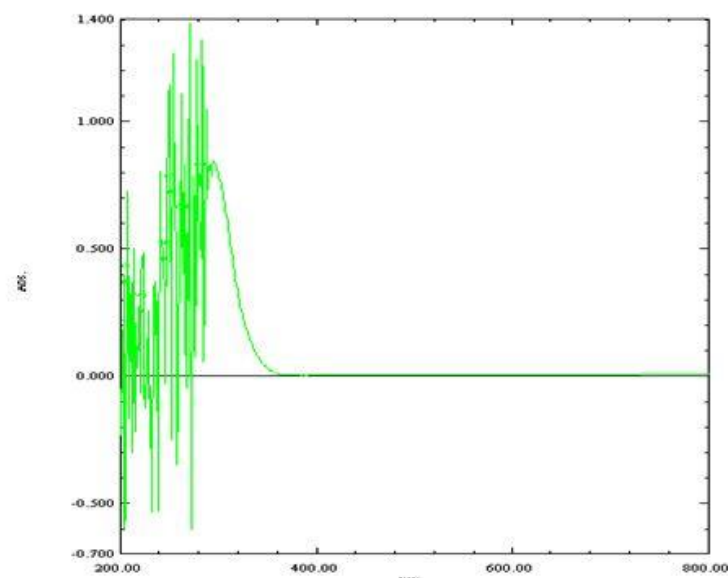


FIG. 9. Absorption spectra of 50 µg/ml of sildenafil citrate pure drug at room temperature showed in the scanned spectral region. The peak at $\lambda=295$ nm was selected as maximum absorbance of the pure SIL because it gives the highest absorption intensity.

TABLE 4. The values of multi-wavelength absorbance of pre-drug of SIL 25 µg/ml, 295 nm was selected as maximum absorbance of the pure SIL because it gives the highest absorption intensity.

No.	Wavelength (nm)	Absorbance
1	295.00	0.418
2	283.00	0.402
3	273.00	0.02
4	265.00	0.152
5	257.00	0.103
6	249.00	0.004
7	239.00	0.332
8	224.00	0.004
9	208.00	-0.047
10	287.00	0.326
11	278.00	-0.02
12	270.00	-0.072
13	260.00	-0.022
14	252.00	-0.105
15	245.00	-0.255
16	229.00	-0.224
17	216.00	-0.22
18	203.00	-0.728

TABLE 5. The values of multi-wavelength absorbance of the drug of SIL 33.33 µg/ml, 295 nm was selected as maximum absorbance of the pure SIL because it gives the highest absorption intensity.

No.	Wavelength (nm)	Absorbance
1	342.00	0.044
2	338.00	0.061
3	323.00	0.173
4	313.00	0.33
5	301.00	0.516
6	299.00	0.537
7	295.00	0.562
8	294.00	0.558
9	292.00	0.538
10	290.00	0.501
11	283.00	0.119
12	282.00	-0.079

TABLE 6. The values of multi-wavelength absorbance of pre-drug of SIL 50 µg/ml, 295 nm was selected as maximum absorbance of the pure SIL because it gives the highest absorption intensity.

S.NO	Wavelength (nm)	Absorbance
1	340.00	0.076
2	331.00	0.151
3	326.00	0.211
4	798.00	0.008
5	315.00	0.434
6	310.00	0.576
7	294.00	0.841
8	299.00	0.797
9	298.00	0.811
10	295.00	0.846
11	292.00	0.782
12	289.00	0.760

Specificity and selectivity

Specificity: Specificity is the ability to assess unequivocally the analyte in the presence of components that may be expected to be present in the sample matrix for demonstrating the specificity of the method for drug formulation [14]. The excipients used in different formulation products did not interfere with the drug peak and thus, the method is specific sildenafil citrate. To further confirm the specificity of the method, UV scans of the spiked drug were taken in the range 200-400 nm and no significant change was found by comparing the absorbance of pure drug and Spiked drug at the analytical wavelength of the drug.

Sensitivity: Following ICH Q2 (R1) recommendations based on the signal to noise approach the Limit of Detection (LOD) was determined for which a signal height to noise ratio of 3 was obtained and for a Limit of Quantitation (LOQ), a signal height to noise ratio of 10 was obtained.

Linearity, LOD, LOQ, and range

Linearity: The linearity of an analytical method is its ability to select test results that are directly, or by a well-defined mathematical transformation, proportional to the concentration of an analyte in samples within a given range. The linearity of the method was observed within the expected concentration range demonstrating its suitability for analysis. The correlation coefficient (r^2) was found to be 0.999 and value of intercept was less than 40 of the response of 100% of the test concentration in all the cases indicating a functional linear relationship between the concentration of the analyte absorbance and sample absorbance (FIG. 10).

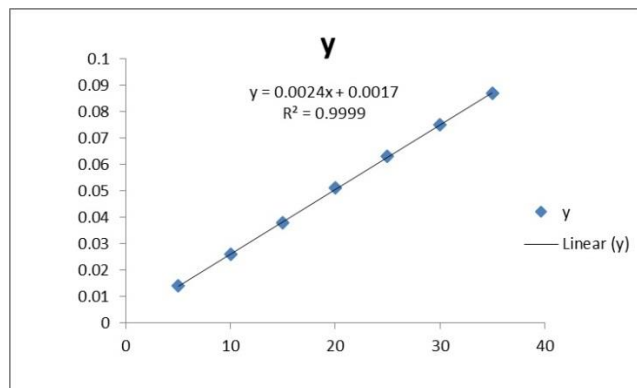


FIG.10. Linearity curve of pure drug (sildenafil citrate).

LOD: The detection limit (LOD) is the lowest amount of an analyte in a sample that can be detected, but not necessarily quantitated, under the stated experimental conditions. It may be expressed as a concentration that gives absorbance value. The lower limit of detection for sildenafil citrate is 2.40 $\mu\text{g/ml}$ in reference material, formulation and 1.60 $\mu\text{g/ml}$ [15].

LOQ: Limit of Quantitation (LOQ) is the lowest amount analyte in a sample that can be determined with acceptable precision and accuracy under the stated experimental conditions. Absorbance value can be taken as LOQ of the method [15]. The LOQ values were found to be less than 2.0 $\mu\text{g/ml}$.

** Limit of detection and quantitation are determined via calculations [16]

** $\text{LOD} = (\text{SD of the response/slope}) \times 3.3$

** $\text{LOQ} = (\text{SD of the response/slope}) \times 10$

Accuracy: The accuracy of an analytical method is the closeness of test results obtained by that method to the true value. In case of the assay of a drug in a formulated product, accuracy may be determined by application of the analytical method to the drug product components to which known amount of analyte has been added within the range of method. If it is not possible to obtain samples of all drug product components, it may be acceptable to add known quantities of the analyte to the drug product (i.e. to spike). In our studies, the later technique was adopted and sildenafil citrate was spiked in the drug product.

Precision: Precision is the degree of reproducibility or repeatability of the analytical method under normal operating conditions. The method passed the test for repeatability as determined by %RSD of the absorbance values of six replicate measurements at 100% test concentration.

Ruggedness: To determine the ruggedness of the proposed method, the test sample solution was analyzed in six replicates comparing %RSD of the measurements by two analysts in the same laboratory.

Robustness: Robustness tested through changing in solvents, temperature, and oxidizing agents which its effect on absorbance (TABLE 7).

TABLE 7. Summary of the present study.

S.No.	Validation parameter	Atorvastatin calcium
1	Detection wave Length	295 nm
2	Beer's Limit	5-40 µg/ml
3	Linearity	5-40 µg/ml
4	R ²	0.9999
5	Intercept	0.0017
6	Slope	0.0024
7	LOD	1.600 µg/ml
8	LOQ	1.85 µg/ml
9	Precision	%RSD<2
10	Recovery	98-102%

Conflicts of Interest

The authors declare no conflicts of interest.

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