

### SYNTHESIS OF 2-ARYLIDENE-4-(SUBSTITUTED ARYL) BUT-3-EN-4-OLIDES AND EVALUATION OF THEIR ANTIBACTERIAL AND ANTI-INFLAMMATORY ACTIVITIES

ASIF HUSAIN\*, M. M. ALAM, M. S. ZAMAN and M. VASEEM ISMAIL

Department of Pharmaceutical Chemistry, Faculty of Pharmacy, Jamia Hamdard (Hamdard University), NEW DELHI - 110 062, INDIA

#### ABSTRACT

Ten new 2-arylidene-4-(substituted aryl)but-3-en-4-olides have been synthesized by condensing  $\beta$ -(4-substituted benzoyl)propionic acid with appropriate aromatic aldehydes in presence of triethylamine (TEA). The compounds have been evaluated for their antimicrobial and anti-inflammatory activities. Their structures were established on the basis of elemental analysis, <sup>1</sup>H NMR and mass spectral data. Some of the compounds were found to have significant activity.

**Key words**:  $\Delta^{\beta,\gamma}$ -Butenolide, Antibacterial, anti-inflammatory.

#### **INTRODUCTION**

Butenolides, consisting of unsaturated  $\gamma$ -lactone ring, are well known heterocycles of biological interest. The  $\gamma$ -lactone ring is a part of variety of natural products like digitalis glycosides, sesquiterpene lactones, lignans and antibiotics<sup>1</sup>. Butenolides and their derivatives are known to have numerous important biological activities<sup>2-5</sup>, which include anti-inflammatory, analgesic, antimicrobial, antitumour, cardiotonic and anticonvulsant. Research from our laboratories and elsewhere has shown that  $\Delta^{\beta,\gamma}$ -butenolides are furnished with antimicrobial and anti-inflammatory actions<sup>6-8</sup>. In continuation of these studies, we hereby report the synthesis, antibacterial and anti-inflammatory activities of ten new  $\alpha$ -arylidene- $\gamma$ -aryl- $\Delta^{\beta,\gamma}$ -butenolides. The compounds were synthesized by following the **Scheme 1** and their structures were established on the basis of elemental analysis, <sup>1</sup>H NMR and mass spectral data.

<sup>\*</sup> Author for correspondence

#### **EXPERIMENTAL**

Melting points were recorded in open glass capillaries and are uncorrected. Microanalyses were performed on Carlo Erba element analyzer and were within  $\pm 0.4\%$  of the theoretical values. <sup>1</sup>H NMR spectra were recorded on Bruker 300MHz instrument in CDCl<sub>3</sub> using TMS as internal standard. Mass spectra of the compounds were recorded on JEOL-DX 303 instrument. Purity of the compounds was checked by TLC on silica gel G coated plates.

## General procedure for the synthesis of $\beta$ -(4-substituted benzoyl) propionic acids (IIa-c)

Succinic anhydride (0.1 mole) was condensed in presence of anhydrous aluminium chloride (0.1125 mole) with appropriate substituted benzenes (50 mL). The reaction mixture was refluxed for four hours and excess solvent was removed by steam distillation. It was purified by dissolving in sodium hydroxide solution (5% w/v), filtered and followed by addition of hydrochloric acid. A solid mass so obtained was filtered, washed with cold water, dried and crystallized from methanol to give **Ha-c** (Table 1).



Scheme 1. Protocol for synthesis of butenolide derivatives (IIIa-j)

# General procedure for the preparation of $\alpha$ -arylidene- $\gamma$ -aryl- $\Delta^{\beta,\gamma}$ -butenolides (IIIa-j)

To a solution of compound II (3 mmol) and appropriate aromatic aldehyde (3 mmol) in acetic anhydride (10 mL), triethylamine (3-4 drops) was added and the reaction mixture was refluxed for four hours under anhydrous condition. After completion of reaction, the mixture was poured onto crushed ice and a coloured solid mass, which separated out, was filtered, washed, dried and crystallized from methanol : chloroform mixture (1 : 1) to give IIIa-j (Table 2).

Comp.	Name of the compound	R	Yield (%)	R <sub>f</sub> value	M. P. (°C )	<sup>1</sup> H NMR (δ-values)
IIa	β-(4-Chloro benzoyl) propionic acid	Cl-	62	0.68	124	2.81 & 3.38 (t, each, 2 x -CH <sub>2</sub> ), 7.45 & 7.92 (d, each, $A_2B_2$ , phenyl).
IIb	β-(4-Ethyl benzoyl) propionic acid	C <sub>2</sub> H <sub>5</sub> -	52	0.7	110	1.25 (t, $3H \underline{CH_3}CH_2$ ), 2.69 (q, $2H CH_3\underline{CH_2}$ ), 2.80 & 3.30 (t, each, 2 x -CH <sub>2</sub> ), 7.27 & 7.85 (d, each, $A_2B_2$ , phenyl).
IIc	β-(4-Methyl benzoyl) propionic acid	CH <sub>3</sub> -	64	0.8	106	2.37 (s, 3H, -CH <sub>3</sub> ), 2.65 & 3.26 (t, each, 2 x -CH <sub>2</sub> ), 7.27 & 7.90 (d, each, A <sub>2</sub> B <sub>2</sub> , phenyl).
s = single	t; $d = doublet; t =$	triplet; a	= quartet	-		

Table 1.P	ivsical and	analvtical	data of f	3-(4-sul	ostituted	benzovl)pr	opionic	acid (	(Ha-c)
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#### Antibacterial activity

The bacterial strains gram positive (*Staphylococcus aureus*) and gram negative (*Eschericia coli*) were used. The test was carried out according to turbidity method<sup>9</sup>. A solution of the compounds was prepared in dimethylformamide and a series of doubling dilutions prepared with sterile pipettes. To each of a series of sterile stoppered test tubes, a standard volume of nutrient broth medium was added. A control tube containing no antimicrobial agent was included. The inoculum consisting of an overnight broth culture of microorganisms was added to separate tubes. The tubes were incubated at 37°C for 24 hours and examined for turbidity. The tube with highest dilution showing no turbidity was the MIC.

#### Anti-inflammatory activity

Carrageenan induced rat paw edema method<sup>10</sup> was employed for evaluating the anti-inflammatory activity of the compounds at a dose level of 20 mg/kg b.w. in albino rats (weighing 100-120 g) using indomethacin as a standard drug for comparison.

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Compd.	R	Ar	Yield (%)	R <sub>r</sub> - value	M.P. (°C )	<sup>1</sup> H NMR (ô-values)	Mass spectra (m/z)
IIIa	CI-		62	0.76	146-48	6.93 (s, 1H, butenolide ring), 7.26 (s, = CH-), 7.44 (m, 5H, H-2,6, p-substituted	282 (M <sup>+</sup> ), 139, 111
						phenyl + H-3,4,5, phenyl), 7.64 (m, 4H, H-3,5, p-substituted phenyl + H-2,6, nhenyl)	
d III	CI-		99	0.78	224-26	5.97 (s, 2H, -OCH <sub>2</sub> O-);, 6.76 (s, 1H,	326 (M <sup>+</sup> ), 139,
						butenolide ring), 7.16 (s, = CH-), 6.84 (s, 1H, H-5, aryl),; 7.08 (m, 2H, H-2,6, aryl),; 7.23 & 7.61 (d, each, A <sub>2</sub> B <sub>2</sub> , p- substituted whenvl)	121
III c	CI-		53	0.8	150-52	7.17 (s, butenolide ring); 7.38 (m, H-2, H. 2' H-3' H-6 H-6' H-7'); 7 97 (m H-	Not taken
						z , 112 , 11-5 , 11-5 , 11-5 , 11-7 ), 12-2 (111, 11- 1 ', H-3, H-4', H-5, H-5', H-8'); 8.34 (s, H-10').	
p III	CI-	OAC	58	0.82	126-28	6.80 (s, butenolide ring), 7.32 (s, = CH-), 7.20 (m 4H arvl), 7.46 & 7.60 (d each	298 (M <sup>+</sup> ), 139
						$A_2B_2$ , p substituted phenyl).	
III e	H <sub>5</sub> C <sub>2</sub> -		56	0.76	176-78	1.19 (t, CH <sub>3</sub> CH <sub>2</sub> ), 2.67 (q, CH <sub>3</sub> CH <sub>2</sub> ); 5.99 (s, -OCH <sub>3</sub> O-); 6.76 (s, butenolide	320 (M <sup>+</sup> ), 133, 105
						ring), 7.19 (s, $=$ CH-); 6.81 (s, H-5'), 7.08 (m, H-2', H-6'), 7.23 & 7.60 (d each, 2x A <sub>2</sub> B <sub>3</sub> , p-substituted phenyl).	
							Cont

Table 2: Physical and analytical data of the  $\alpha$ -arylidene- $\gamma$ -aryl- $\Delta^{\beta,\gamma}$ -butenolides (IIIa-j)

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Table 2: P	hysical and	I analytical data of the	a-arylide	ene-γ-aryl-∆	<sup>β,γ</sup> -butenolid	es (IIIa-j)	
Compd.	R	Ar	Yield (%)	R <sub>r</sub> value	M.P. (°C )	<sup>1</sup> H NMR (ð-values)	Mass spectra (m/z)
Шf	H <sub>5</sub> C <sub>2</sub> -	C	68	0.72	140-42	1.19 (t, CH <sub>3</sub> CH <sub>2</sub> ), 2.62 (q, CH <sub>3</sub> CH <sub>2</sub> ); 6.76 (s, butenolide ring), 7.23 (s, = CH-), 7.20 (m, H-3 ', H-5 '); 7.48 (m, H-2 ', H- 6'); 7.36 & 7.62 (d each, 2x A <sub>2</sub> B <sub>2</sub> , p- substituted phenyl).	310 (M <sup>+</sup> ), 133, 105
III	H <sub>5</sub> C <sub>2</sub> -	N <sup>2</sup> O	62	0.75	96-98	1.29 (t, CH <sub>3</sub> CH <sub>2</sub> ); 2.72 (q, CH <sub>3</sub> CH <sub>2</sub> ); 6.89 (s, butenolide ring); 7.4 (s, = CH-); 7.67 (m, H-5 '); 7.29 & 8.25 (d, H-4', H- 6'); 8.50 (s, H-2'); 7.32 & 7.73 (d each, 2x A <sub>2</sub> B <sub>2</sub> , p-substituted phenyl).	321(M <sup>+</sup> ), 133, 105
Ч Ш	H <sub>5</sub> C <sub>2</sub> -		48	0.88	156-58	1.20 (t, CH <sub>3</sub> CH <sub>3</sub> ); 2.6 (q, CH <sub>3</sub> CH <sub>2</sub> ); 7.0 (s, butenolide ring); 7.18 (s, = CH-); 7.39 (m, H-2', H-2, H-3', H-6', H-6, H-7'); 7.9 (m, H-1', H-3, H-4', H-5', H-5, H- 8'); 8.35 (s, H-10').	Not taken
Ĩ	H <sub>5</sub> C <sub>2</sub> -	CH <sub>3</sub>	56	0.76	119-21	1.19 (t, CH <sub>3</sub> CH <sub>2</sub> ); 2.59 (q, CH <sub>3</sub> CH <sub>2</sub> ); 2.34 (s, CH <sub>3</sub> ); 6.80 (s, butenolide ring); 7.31 (s, = CH-); 7.19 (m, H-2', H-3', H- 4', H-5'); 7.46 & 7.61 (d each, 2x A <sub>2</sub> B <sub>2</sub> , p-substituted phenyl).	290 (M <sup>+</sup> ), 133, 105
Ϊ	H <sub>3</sub> C-		65	0.75	86-88	2.4 (s, CH <sub>3</sub> ); 6.73 (s, butenolide ring); 7.38 (s, = CH-), 7.30 & 7.41 (d each, $2x$ A <sub>2</sub> B <sub>2</sub> , p-substituted phenyl), 7.48 (m, 5H, phenyl),.	262 (M <sup>+</sup> ), 119
s = singlet;	; d = double	t; t = triplet; q = quatret	;; m = mul	ltiplet			

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The percentage inhibition of inflammation was calculated by applying Newbould formula<sup>11</sup>. The ulcerogenic activity was carried out by the reported method<sup>12</sup>. None of these compounds showed ulcerogenic activity.

Antibacterial activity (Minimum inhibitory concentration*)		Anti-inflammatory activity (% inhibition in rat paw oedema)				
Comp.	S. aureus	E. coli	Normal paw volume (x)	Paw oedema 3 hr after carragenan (a)	% inhibition (1- a-x/b-y) 100 of oedema	
IIIa	>100	>100	$0.70\pm0.03$	$0.93\pm0.03$	25.80	
IIIb	-	-	$0.65\pm0.03$	$0.92\pm0.02$	12.90	
IIIc	25	50	$0.73\pm0.03$	$0.95\pm0.04$	29.03	
IIId	50	25	$0.68\pm0.03$	$0.81\pm0.03$	58.06	
IIIe	-	-	$0.65\pm0.03$	$0.90\pm0.02$	19.35	
IIIf	25	12.5	$0.73\pm0.03$	$0.95\pm0.04$	29.03	
IIIg	>100	>100	$0.72\pm0.02$	$0.86\pm0.03$	54.83	
IIIh	10	25	$0.71\pm0.04$	$0.96\pm0.05$	19.35	
IIIi	>100	-	$0.71\pm0.02$	$0.88\pm0.03$	45.16	
IIIj	-	>100	$0.67\pm0.03$	$0.93\pm0.04$	16.12	
Nitrofurazone 12.5 6.5						
Indomethacin		$0.73\pm0.03$	$0.85\pm0.03$	61.30		
Control			$0.68 \pm 0.02$ (y)	$0.99 \pm 0.03(b)$		

#### Table 3. Biological activity

\* = in  $\mu$ g/mL; - = insignificant activity

#### **RESULTS AND DISCUSSION**

The newly synthesized butenolides showed interesting profile of antibacterial and anti-inflammatory activity. In antibacterial test, compound **IIIf** showed very good activity

against *E. coli* (MIC-12.5  $\mu$ g/mL) and good activity against *S. aureus* (MIC-25  $\mu$ g/mL). In anti-inflammatory test, the most active compound was **IIId**, which showed 58.06 % inhibition comparable with the standard drug indomethacin (61.3 %) at a dose level of 20 mg/kg body weight. It is significant that none of these compounds showed ulcerogenic activity, which is a common feature with NSAIDs.

#### ACKNOWLEDGEMENT

The authors are thankful to AICTE (RPS-Scheme) for financial support. We also thank Dr. S. Shah, Incharge, Animal House, Jamia Hamdard, for providing animals for activity.

#### REFERENCES

- 1. F. Andrew, F. Elizabeth and G. Anne-Geraldine, *PCT Int. Appl. WO*, 01, 44, 221 (2001).
- 2. V. O. Kozminykh, N. M. Igidov, E. N. Kozminykh, Z. N. Semenova and Y. Andreichikov, Pharmazie, **47**, 261 (1992).
- 3. Y. Satomi, H. Nishino, A. Iwashima, M. Posrihara, Y. Tamai and M. Ito, Anticancer Drug Des., 7, 169 (1992).
- 4. E. Dacre and K. Staffan,, PCT Int. Appl. WO 01, 43, 739 (2001).
- 5. L. Eric, K. Derek, S. Harjit, M. Isidro, B. Adiba, A. Washington and J. T. Michael, Pharm. Pharmacol. Lett., **11**, 05 (2001).
- A. Husain, S. M. Hasan, A. Kumar and M. M. Alam, Asian J. Chem., 17, 1579 (2005).
- 7. A. Husain, S. M. Hasan, S. Lal and M. M. Alam, Ind. J. Hetr. Chem., 14, 163, (2004).
- 8. C. K. Lau et al., Biorg. Med. Chem. Lett., 15, 3187 (1999).
- 9. R. Cruickshank, J. P. Dugid, D. P. Marmion and R. H. A. Swain, Medical Microbiology, Vol. II, Churchill-Livingstone, London (1975), p. 2.
- 10. C. A. Winter, E. A. Risley and G. N. Nuss, Proc. Soc. Exp. Biol., 111, 544 (1962).
- 11. B. B. Newbould, Brit. J. Pharmacol., **21**, 157 (1963).
- 12. G. Wilhemi and R. Menass-Gydnia, Pharmacology, 8, 321 (1972).