



Synthesis, characterization and antimicrobial screening of 3-(1H-benzo[d]imidazol-2-ylsulfanyl) methyl-4-[phenoxy(phenyl)acetamido]-5-mercapto-1,2,4-triazole and related aryloxy compounds

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ABSTRACT

A series of new 1H-benzo[d]imidazole derivatives of 3,4-substituted triazole. 3-(1H-Benzo[d]imidazol-2-ylsulfanyl)methyl-4-[phenoxy(phenyl)acetamido]-5-mercapto-1,2,4-triazole and related aryloxy compounds were synthesised, analysed and characterised by FTIR, ¹HNMR and elemental analysis. These compounds were screened for antibacterial and antifungal activity. The antibacterial activities were compared against chlorophenicol and antifungal activity with mycostalin. Some triazole derivatives showed positive antibacterial and antifungal activity.

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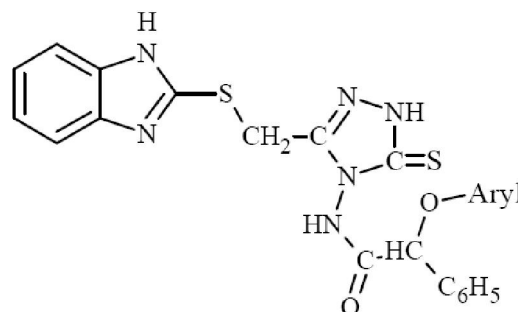
KEYWORDS

Synthesis;
 Microbial activity;
 Benzo[d]imidazol-2-ylsulfanyl methyl triazole derivatives.

INTRODUCTION

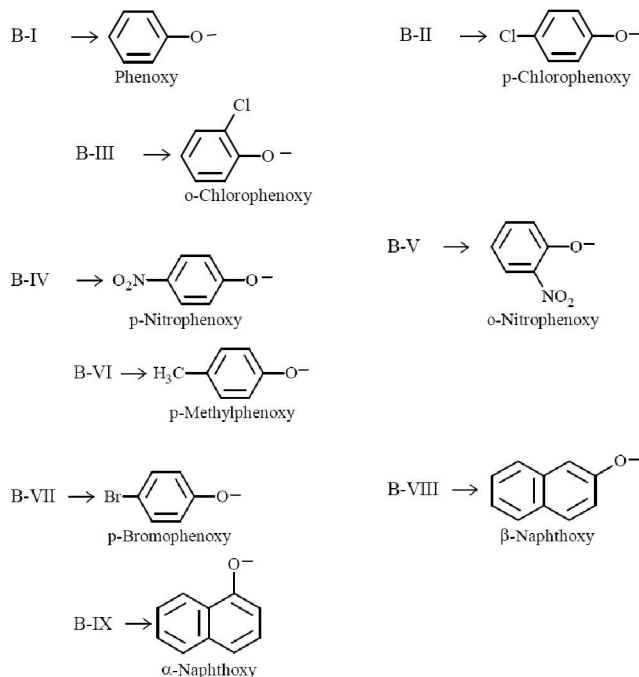
A perusal of literature revealed that benzimidazole and triazole ring containing heterocyclic molecules possess wide range of antibacterial activity and medicinal properties^[1-10]. In addition, triazolobenzimidazoles possess broad range of antimicrobial spectrum and have privileged nuclei to display medicinal activity. Benzimidazole derivatives possess great importance in medicinal chemistry due to wide variety of pharmacological activity^[11,12] in controlling cardiovascular diseases^[13], anticancer properties^[14], antiinflammatory^[15], antibacterial^[16], antifungal^[17], antidiabetic^[18] and anti HIV^[19] activity and some benzimidazoles are antioxidant^[20-24]. It has also been noticed that certain benzimidazole derivatives like ciprofloxacin and norfloxacin are most essential and popular antibiotics. In view of substantial pharmacological importance and multidimensional applications of benzimidazole-triazole mixed heterocyclic

compounds we were motivated to study their chemistry and here we report synthesis, characterisation and antibacterial activity of new substituted triazole ring containing 1H-benzo[d]imidazole-2ylsulfanylmethyl derivatives, 3-(1H-Benzo[d]imidazole-2-ylsulfanyl)methyl-4-[phenoxy(phenyl)acetamido]-5-mercapto-1,3,4-triazole (BSPT) and its nine related aryloxy derivatives (B-I to B-IX).



BSPT
 (B-I to B-IX)

Aryloxy-group (-O-Aryl) for BI– B-IX



EXPERIMENTAL

All the reagents and chemicals were obtained from E Merck, Lobachem, Chem pure, Sigma Aldrich and Fluka (Germany). Solvents used for synthesis were analytical grade reagent. The purity of the products was checked by TLC. The purity of known and reported chemicals were ascertained from MP and estimation of nitrogen. The FTIR spectra of compounds were recorded in KBr disc on Shimadzu, IR Spectrophotometer-2500. The ¹HNMR spectra of compounds were recorded on a Bruker AV-400 spectrophotometer in CDCl₃ or DMSO or DMF-d₆. The CHNS analysis reports, were obtained from CDRT Lucknow or BIT Mesra, Ranchi.

The compound BSPT (BI-B-IX) were synthesized using

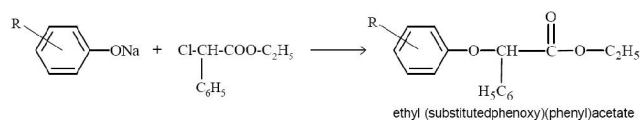
- Aryloxyphenylacetic acid hydrazide (A-I to A-IX)
- Ethyl bromoacetate
- 1H-Benzo[d]imidazole-2-ylthiol
- Potassium dithiocarbazinate of (1H-benzo[d]imidazole-2-ylthio)methylcarbohydrazide, adopting scheme A, B and C
- Phenoxy(phenylacetic acid) hydrazide and its derivatives were synthesised using Scheme A.
- The compound 'b' and 'c' were obtained from

market and they were Fluka Product. Both b and c were used without further purification.

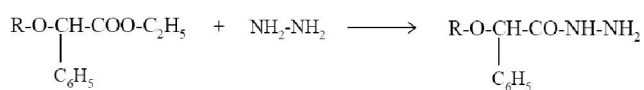
- Potassium dithiocarbazinate of (1H-benzo[d]imidazole-2-ylthio)methylcarbohydrazide was prepared adopting Scheme B.

Preparation of aryloxy phenyl acetic acid hydrazide

Sodium salt of phenol and substituted phenols were refluxed with ethyl (phenyl chloroacetate) [C₆H₅-CHCl-COOC₂H₅] in dioxane on steam bath for 3 hours and the crude aryloxy compound aryloxy ethyl (phenylacetate) obtained above was refluxed with 98% hydrazine hydrate on a steam bath for 3-4 hours. The product formed was titrated with ether to remove unreacted phenol and ester. The white mass left was recrystallised with aqueous ethanol. The related substituted aryloxyphenylacetic acid hydrazides (A-I to A-IX) were also prepared following the above procedure.



The reaction of aryloxy ester (RO-CH(C₆H₅)-COOC₂H₅) with hydrazine takes place as shown below:-



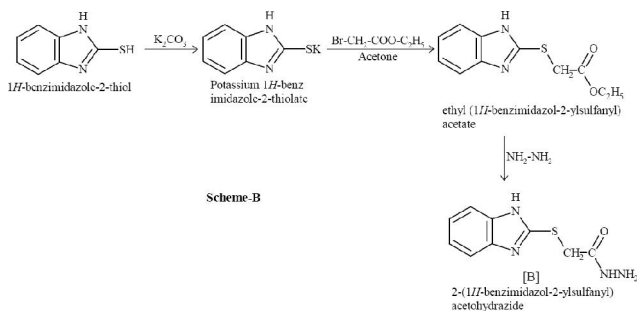
The melting point and analytical results of acetohydrazide A-I to A-IX are given in TABLE 1

Preparation of (1H-benzo[d]imidazol-2-ylsulfanyl)acetic acid hydrazide, from 1H-benzo[d]imidazole-2-thiol.

Potassium salt of 1H-Benzo[d]imidazole-2-thiol (BtH) was prepared by heating aqueous ethanol solution of thiol (BtH) with calculated amount of K₂CO₃ and the potassium 1H-benzo-[d]-imidazol-2-thiolate (KBt) was obtained by evaporating the solution to dryness. The dried product was suspended in dry acetone and refluxed with ethylbromoacetate (Br-CH₂COOC₂H₅) with stirring. The resulting solution was filtered and solvent evaporated to get solid product. The crude ester obtained was refluxed with 98% hydrazine hydrate in 30 ml THF to yield 2-[1H-

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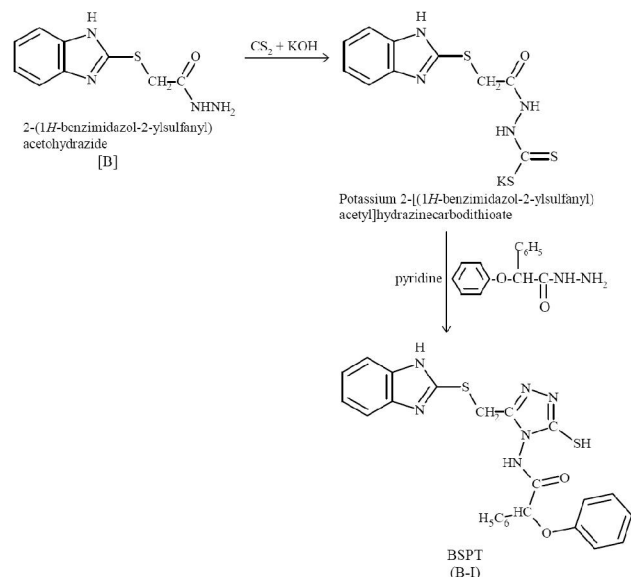
benzo[d]imidazol-2-ylsulfanyl)acetohydrazide (M.P-236-237°C Reported-236°C)



The purity of products were ascertained from TLC, melting point and C, H, N analysis of recrystallised product (B).

The triazole containing benzimidazole derivatives (B-I to B-IX) were obtained by refluxing potassium dithiocarbazinate of (1H-benzo[d]imidazol-2-ylthio)methylcarbonohydrazide and aryloxy(phenylacetohydrazide in pyridine as given in Scheme C

2-(1H-benzo[d]imidazol-2-ylsulfanyl)acetohydrazide was suspended in ethanol-dioxane mixture and treated with calculated amount of CS₂ and KOH and stirred for two hour to get potassium salt of dithiocarbazinate of B. Potassium salt of dithiocarbazinate of 'B' was refluxed with aryloxy(phenylacetic acid)hydrazide in pyridine when pyridinium salt of 3-(1H-benzo[d]imidazole-2-ylsulfanyl)methyl-4-[aryloxyphenylacetamido]-5-mercapto-1,3,4-triazole [BSPT] was obtained. The free mercaptotriazole derivatives (B-I to B-IX) were obtained by neutralising the product with dilute hydrochloric acid.



The compounds B-II to B-IX were obtained by the same procedure using appropriate aryloxy group. The analytical results of product B-I to B-IX are given in TABLE 2. The i.r spectral data of products B-I to B-IX and A-I to A-IX are given in TABLE 3 and 4

EXPERIMENTAL

General method for preparation of A-I to A-IX

10 millimoles of phenol or substituted phenol or naphthols was taken in 50 ml dioxan or THF and refluxed with calculated amount of KOH to get potassium salt of phenol. The potassium phenolate was treated with 10 millimole of ethylphenylchloroacetate and refluxed gently for three hours and solid separated (KCl) was removed by filtration. The filtrate containing ethyl (aryloxy phenyl acetate) was treated with hydrazine hydrate (98%) and refluxed gently for three to four hours on steam bath and solvent evaporated to get syrupy mass. The product on cooling gave solid product which was recrystallised with hot aqueous ethanol or tetrahydrofuran (yield 80-85%).

The method of preparation of B

About 0.1 mole of potassium salt of (1H-benzo[d]imidazol-2-thiol) was taken in dry acetone and refluxed on steam bath with 0.1 mole of ethylbromoacetate with stirring for 3 hours. The sulfanylacetate was obtained on evaporation of acetone. The product was collected and recrystallised with etha-

TABLE 1 : Elemental analysis of compound A-I to A-IX

| S.N | Molecular formula | M.P °C | % analysis, found (Calculated) | | |
|--------|--|------------|--------------------------------|-------------|------------------|
| | | | C | H | N |
| A-I | C ₁₄ H ₁₄ N ₂ O ₂ | 243-242 | 69.65 (69.42) | 6.03 (5.78) | 11.71 ?? (11.57) |
| A-II | C ₁₄ H ₁₃ N ₂ O ₂ Cl | 257-216.54 | 60.06 (60.75) | 4.66 (4.70) | 10.01 (10.12) |
| A-III | C ₁₄ H ₁₃ N ₂ O ₂ Cl | 249-216.54 | 60.83 (60.75) | 4.83 (4.70) | 10.23 (10.12) |
| A-IV | C ₁₄ H ₁₃ N ₃ O ₄ | 262-287 | 58.17 (58.53) | 4.36 (4.52) | 14.43 (14.63) |
| A-V | C ₁₄ H ₁₃ N ₃ O ₄ | 260-287 | 58.41 (58.53) | 4.63 (4.52) | 14.63 ? |
| A-VI | C ₁₅ H ₁₆ N ₂ O ₂ | 241-256 | 70.13 (70.31) | 6.11 (6.25) | 10.83 (10.93) |
| A-VII | C ₁₄ H ₁₃ N ₂ O ₂ Br | 248-312 | 52.16 (52.33) | 3.98 (4.05) | 8.69 (8.72) |
| A-VIII | C ₁₈ H ₁₆ N ₂ O ₂ | 258-292 | 73.73 (73.79) | 5.61 (5.48) | 9.39 (9.59) |
| A-IX | C ₁₈ H ₁₆ N ₂ O ₂ | 263-264 | 73.68 (73.97) | 5.53 (5.48) | 9.41 (9.59) |

nol tetrahydrofuran mixture. The acetate was refluxed with 98% hydrazide-hydrate to get 2-(1H-benzo[d]imidazol-2-yl) sulphonylacetylhydrazide.

General procedure for preparation of B-1 to B-IX

About 10 millimole of B was taken in 30 ml ethanol and calculated amount of 10-12 millimole CS₂ and 10 millimole KOH were mixed and refluxed with stirring for 2-hour when potassium salt of dithiocarbamate separated. The product was dissolved in 20 ml pyridine and refluxed with 10 millimoles of aryloxyphenyl acetic acid hydrazide [Aro-CH(Ph)CO NH-NH₂] for 3–4 hours. The refluxate on cooling gave cream yellow crystalline precipitate of pyridinium salts of B-I to B-IX. The free mercaptotriazole was obtained by suspending the pyridinium salt in 30-40 ml water and neutralising it with dilute HCl. The free triazole was recrystallised with ethanol-THF mixture. The result of elemental analyses and mpt of B-1 to B-IX are given in TABLE 2.

TABLE 2 : Elemental analysis of compound B-I to B-IX

| S.N | Molecular formula | M.P °C | % analysis, found (Calculated) | | | |
|-----------------|---|--------|--------------------------------|----------------|------------------|------------------|
| | | | C | H | N | S |
| B-I 488.3 | C ₂₄ H ₂₀ N ₆ O ₂ S ₂ | 276-77 | 58.78 (58.99) | 4.12 (4.09) | 17.32 (17.21) | 13.26 (13.10) |
| B-II 522.84 | C ₂₄ H ₁₉ N ₆ O ₂ S ₂ Cl | 281-82 | 55.12 (55.08) | 3.76 (3.63) | 15.94 (16.06) | 12.13 (12.24) |
| B-III 522.84 | C ₂₄ H ₁₉ N ₆ O ₂ S ₂ Cl | 280-81 | 54.93 (55.08) | 3.72 (3.63) | 15.89 (16.06) | 12.16 (12.24) |
| B-IV 533.4 | C ₂₄ H ₁₉ N ₇ O ₄ S ₂ | 294-95 | 54.13 (54.03) | 3.61 (3.56) | 18.26 (18.37) | 11.86 (12.00) |
| B-V 533.4 | C ₂₄ H ₁₉ N ₇ O ₄ S ₂ | 296-97 | 54.01 (54.03) | 3.51 (3.56) | 18.15 (18.37) | 11.92 (12.00) |
| B-VI 502.3 | C ₂₅ H ₂₂ N ₆ O ₂ S ₂ | 284-85 | 59.58 (59.73) | 4.46 (4.38) | 16.58 (16.72) | 12.64 (12.74) |
| B-VII 568.2 | C ₂₄ H ₁₉ N ₆ O ₂ S ₂ Br | 291-92 | 50.63 (50.68) | 3.28 (3.34) | 15.21 (15.13) | 11.16 (11.26) |
| B-VIII 538.2 | C ₂₈ H ₂₂ N ₆ O ₂ S ₂ | 288-84 | 64.32 (64.46) | 4.12 (4.08) | 15.51 (15.60) | 11.68 (11.89) |
| B-IX | C ₂₈ H ₂₂ N ₆ O ₂ S ₂ | 290-91 | 64.14 (64.46) | 4.21 (4.08) | 15.71 (15.60) | 11.81 (11.89) |

RESULT AND DISCUSSION

The ¹HNMR spectrum of A-I shows a singlet at $\delta = 3.68$ ppm for (-O-CH(C₆H₅)-CO) for aceto(C-H) proton. The phenyl ring CH proton signals were observed between $\delta = 6.965$ and 7.835 ppm as multiplet and broad NH, NH₂ proton signals at $\delta = 5.315$ - 5.685 ppm.

The ¹HNMR spectrum of A-II shows a singlet at 3.726 ppm for -O-CH(C₆H₅)CO and phenyl proton signals as multiplet between $\delta = 7.015$ – 7.935 ppm

and a broad NH, NH₂ proton signals were located at 5.465 – 5.285 ppm.

The p-nitrophenoxy derivative A-III shows a signal at 3.865 ppm and phenyl ring proton signals between $\delta = 7.115$ – 8.215 ppm as multiplet. The NH and NH₂ proton signals was broad at $\delta = 5.415$ – 5.845 ppm.

The β -naphtholxy compound A-VIII shows a singlet at $\delta = 3.745$ ppm attributed from (-O-CH-CO) proton and phenyl ring C—H proton signals as multiplet between $\delta = 7.015$ and 7.895 ppm. The NH and NH₂ proton signals were observed between $\delta = 5.425$ - 5.845 ppm.

The ¹HNMR spectra of hydrazides A-I to A-IX are consistent with proposed structure and these are supported by FTIR and elemental analysis. The i.r spectral band positions of A-I to A-IX and triazole product B-I to B-IX are recorded in TABLE 3 & 4.

The i.r spectra of phenoxy and related aryloxy (phenylacetic acid)hydrazide show characteristic NH₂, NH, phenyl C-H and amido CO stretches in 3μ - 16μ region and NH₂, NH, C-H and CO stretches as well as NH₂ bending and phenoxy Ph-O-C stretches of compounds A-I to A-IX are recorded in TABLE 3. The NH₂ and NH stretches were located between 3348 - 3105 cm⁻¹ and phenyl ring (C—H) stretches between 3085 - 3050 cm⁻¹. The strong band located at 1685 - 1698 cm⁻¹ is assigned to ν (CO) of amide group. The medium band located near 1636 - 1628 cm⁻¹ is attributed to δ (NH₂) of hydrazide group (CO-NH-NH₂). A medium band located at 1063 to 1050 cm⁻¹ (TABLE 3) is attributed

TABLE 3 : Prominent I.R bands in cm⁻¹ of compounds A-I to A-IX

| Compound | ν NH ₂ , ν NH, ν C—H | ν CO | δ NH ₂ | ν (C-O-C) |
|----------|---|----------|--|---------------|
| A-I | 3335, 3220, 3105, 3070 | 1698 | 1634 | 1065 |
| A-II | 3325, 3218, 3140, 3082 | 1692 | 1628 | 1055 |
| A-III | 3340, 3241, 3165, 3075 | 1685 | 1632 | 1054 |
| A-IV | 3316, 3205, 3148, 3070 | 1690 | 1636 ν (NO ₂) 1483 1630 | 1056 |
| A-V | 3345, 3233, 3140, 3060 | 1688 | 1633 ν (NO ₂) 1481 | 1050 |
| A-VI | 3340, 3231, 3150, 3073, 2982, 2860 | 1695 | 1633 | 1062 |
| A-VII | 3348, 3268, 3140, 3065 | 1690 | 1638 | 1061 |
| A-VIII | 3340, 3240, 3165, 3069 | 1687 | 1636 | 1063 |
| A-IX | 3335, 3245, 3160, 3085 | 1698 | 1630 | 1058 |

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to aryloxy (C—O—C) stretching vibration. These i.r bands of compound A-I to A-IX are consistent with proposed structure of aryloxy(phenylacetic acid)hydrazide.

The prominent IR band due to $\nu(\text{NH})$, ring $\nu(\text{NH})$, $\nu(\text{CH}_2)$, $\nu(\text{C-H})$, phenyl ring, $\nu(\text{C=S})$, $\nu(\text{CO})$, $\delta(\text{NH})$ etc were consistent with proposed structure of triazole derivatives and are recorded in TABLE 4. The i.r spectrum of B-I, 3-[2-(1H-benzo[d]imidazol-2-ylsulfanyl)methyl]-4-[phenoxy(phenylacetamido)]-5-mercapto-1,2,4-triazole shows NH and (C-H) stretching vibrations at 3265, 3105, 2940 and 2865 cm^{-1} (TABLE 4)

The i.r spectra of all benzimidazole derivatives B-I to B-IX show strong $\nu(\text{CO})$ vibration between 1685-1705 cm^{-1} confirming the presence of acetamide (-CONH) group. The ring NH and amide NH stretches were observed as medium band between 3265-3105 cm^{-1} . The $-\text{CH}_2-$ stretches of sulfanyl methyl ($-\text{S-CH}_2-$) and acetamido ($-\text{CO-CH-}$) group were located at 2860-2940 cm^{-1} . The I.R bands at 1590-1610 cm^{-1} observed in B-I to B-IX are assigned to ring (C=N) stretching vibrations. The nitro aryloxy compound B-IV and B-V show NO_2 band at 1481-1483 cm^{-1} .

The $\nu(\text{S-H})$ attached to triazole ring could not be observed indicated the predominance of thione tautomer in the molecule. The $\delta(\text{NH})$ of B-I to B-IX were observed between 1526-1508 cm^{-1} and $\nu(\text{C=S})$ band could be assigned to a strong i.r band observed between 1342-1305 cm^{-1} . The phenoxy ($-\text{C-O-C-}$) stretch can be assigned to a medium i.r band near 1020 ± 10 cm^{-1} . A large number of IR band located in finger print region are assigned to phenyl and triazole ring skeletal vibrations.

The proton NMR spectrum of B-I shows ($-\text{S-CH}_2$) proton signal at $\delta = 2.281$ ppm as singlet as well as acetamide $-\text{CH-}$ proton at $\delta = 3.965$ ppm as singlet. The ring NH proton signal was observed at 8.652 ppm, 8.925 ppm. The phenyl proton signals were observed at $\delta = 7.055$ -7.985 ppm as multiplet.

The $^1\text{HNMR}$ spectrum of chloro, bromo and nitro-aryloxy derivatives are almost identical. The phenyl proton signals were located between 6.943-7.855 ppm and (S-CH_2) proton as singlet between 2.835-2.945 ppm.

The (S-CH_2) proton of B-II were observed at $\delta =$

2.865 ppm and acetoxy ($\text{CO-CH}(\text{C}_6\text{H}_5)\text{-O-}$) proton signal at $\delta = 3.685$ ppm. Its phenyl proton signals were observed as multiplete between $\delta = 7.154$ -7.925 ppm. The rings NH of triazole and benzimidazole proton signals were located at 8.765 and 9.254 ppm. The acetamido (NH) proton signal were located at 5.45 ppm as singlet as broad band.

The proton NMR of 3-[2-(1H-benzo[d]imidazole-2-ylsulfanyl)methyl]-4-[(p-methylphenoxy(phenylacetatamido)]-5-mercapto-1,2,4-triazole (B-VI) displays $-\text{CH}_3$ proton signals at $\delta = 1.695$ ppm as singlet. The $-\text{S-CH}_2-$ proton signal was observed as singlet at $\delta = 2.945$ (2H, $-\text{S-CH}_2-$) and $-\text{O-CH-}$ (C_6H_5)CO proton signal at $\delta = 3.875$ ppm. The broad singlet at 5.45 ppm was assigned to acetamide ($-\text{HN-CO-CH-}$) proton signal. The phenyl ring (C-H) proton signals were located between $\delta = 7.025$ -7.845 ppm. The ring NH proton signals were located as singlet at 8.735 and 8.952 ppm. Based on spectral data of the compound, the structures suggested for the synthesized product (BSPT) were also supported by the analytical compositions of benzimidazole derivatives.

Antibacterial and antifungal activity

The antifungal activity of BSPT (Compound B-I to B-IX) were evaluated by radical growth method^[26] using Czepek agar medium prepared by dissolving 20 g starch, 20 g agar 20 g glucose in one litre distilled water. The resulting solution was added requisite amount of test compound to get 100 and 200 ppm solution. The medium was then poured into petri plate and spore of fungi were placed on medium with the help of inoculum needle. These petri plate were wrapped in polythene bags by mixing 2 drops of ethanol and placed in an incubator at $30 \pm 0.5^\circ\text{C}$. The linear growth of fungi was calculated by measuring the fungal colony diameter after five days. The percentage inhibition was calculated

using the relation $\frac{C - T}{C} \times 100$, where C & T are the

diameter of the fungus colony and control test plate respectively. The fungi used in present microbial screening are candida albicans, F oxysporum, Aspergillusflavus, R. phaseoli and A.nigar. The control solution was mycostalin. The result of activity shown in TABLE 5. Almost all benzimidazolylsulfanylmethyltriazole derivatives have

causes inhibition of fungal growth but the activity of nitrophenoxy derivatives (B-IV & B-V) were quite encouraging comparable to mycostalin. The activity of B-I to B-IX were larger with candida albicans and A nigar.

The antibacterial activity against E.coli, S.aureus and Bacillus subtilis were studied for compounds B-I to B-IX and zone of inhibition was observed in all the derivatives. The activity was studied by zone inhibition technique^[27]. The nutrient agar medium was prepared by dissolving 5 g peptone 5 g beef extract, 5 g NaCl

and 20 g agar agar in one litre distilled water. The medium solution was pipetted into petri plate and dried, the dried plate was seeded with bacteria and test compound dissolved in DMF (250 ppm 500 ppm strength). The disc of whatmanfilterpaper soaked with these solutions to 5 mm diameter discs were dried and placed on medium previously soaked with organism in petriplate at suitable distance and incubated at $30 \pm 1^\circ\text{C}$ for 24 hours. The zone of inhibition was measured accurately in mm. The results of inhibition are recorded in TABLE 5. It was encouraging to note that compounds

TABLE 4 : Diagnostics IR bands of compounds (B-I to B-IX) in cm^{-1}

| S.No | $\nu(\text{NH})$ | $\nu(\text{CH}_2) + \nu(\text{C-H})$ | $\nu(\text{CO})$ | $\nu(\text{C=N})$ | $\delta(\text{NH})$ | $\nu(\text{C=S})$ | $\nu(\text{C-O-C})$ |
|--------|------------------|--------------------------------------|------------------|-------------------|---------------------|-------------------|---------------------|
| B-I | 3265, 3105 | 2865, 2940 | 1695 | 1601 | 1512 | 1320 | 1024 |
| B-II | 3260, 3132 | 2950, 2840 | 1690 | 1608 | 1508 | 1312 | 1015 |
| B-III | 3246, 3135 | 2940, 2830 | 1685 | 1595 | 1521 | 1342 | 1018 |
| B-IV | 3211, 3130 | 2955, 2845 | 1692 | 1598 | 1526 | 1338 | 1022 |
| B-V | 3245, 3136 | 2962, 2842 | 1690 | 1605 | 1522 | 1320 | 1030 |
| B-VI | 3256, 3151 | 2960, 2841 | 1696 | 1601 | 1513 | 1324 | 1015 |
| B-VII | 3220, 3140 | 2965, 2845 | 1700 | 1605 | 1508 | 1305 | 1028 |
| B-VIII | 3245, 3160 | 2932, 2890 | 1685 | 1602 | 1518 | 1321 | 1021 |

TABLE 5 : Antibacterial and antifungal activity of compounds B-I to B-IX; Antifungal inhibition after 5 days and antibacterial inhibition after 24 hrs.

| Fungi or bacteria | Conc in ppm | B-I | B-II | B-III | B-IV | B-V | B-VI | B-VII | B-VIII |
|-------------------|-------------|-----|------|-------|------|-----|------|-------|--------|
| A.flavus | 100 ppm | 30 | 25 | 30 | 32 | 45 | 46 | 16 ? | 25 |
| | 200 ppm | 42 | 38 | 45 | 46 | 55 | 56 | 32 | 30 |
| Candida albicans | 100 ppm | 55 | 50 | 52 | 48 | 53 | 36 | 30 | 35 |
| | 200 ppm | 70 | 68 | 68 | 50 | 68 | 50 | 35 | 40 |
| F oxysporum | 100 ppm | 45 | 38 | 39 | 40 | 40 | 38 | 30 | 41 |
| | 200 ppm | 56 | 49 | 52 | 48 | 52 | 52 | 36 | 48 |
| R.Phaseoli | 100 ppm | 48 | 45 | 40 | 42 | 50 | 33 | 25 | 22 |
| | 200 ppm | 60 | 56 | 53 | 50 | 56 | 48 | 32 | 43 |
| A Nigar | 100 ppm | 42 | 45 | 50 | 52 | 54 | 50 | 35 | 40 |
| | 200 ppm | 55 | 60 | 62 | 64 | 68 | 62 | 48 | 52 |

Antibacterial activity

| | Conc. in ppm | B-I | B-II | B-III | B-IV | B-V | B-VI | B-VII | B-VIII | Reference |
|------------|--------------|-----|------|-------|------|-----|------|-------|--------|-----------|
| E.Coli | 250 | 6 | 5 | 6 | 8 | 8 | 5 | 3 | 3 | 22a |
| | 500 | 8 | 7 | 8 | 14 | 15 | 7 | 5 | 5 | 24a |
| S.aureus | 250 | 4 | 4 | 5 | 10 | 9 | 6 | 4 | 4 | 22a |
| | 500 | 7 | 6 | 7 | 14 | 13 | 8 | 6 | 7 | 24 a |
| B Subtilis | 250 | 5 | 5 | 6 | 11 | 10 | 5 | 5 | 5 | 23 b |
| | 500 | 7 | 8 | 8 | 14 | 14 | 9 | 7 | 8 | 26 b |

Standard for antifungal growth, Mycostaline; Standard for antibacterial activity was ciprofloxacin(a) Streptomycin(b).

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were highly active on *Escherichia coli*. The standard used was chlorophenicol.

The infrared and ¹HNMR spectral data of compound A-I and A-IX

A-I Phenoxy (phenylacetic acid)hydrazide or aryloxyphenylacetic acid hydrazide show prominent IR bands for NH₂, NH, C-H, aromatic C-H, amido (CO), phenoxy (C—O—C) stretches and phenyl ring skeletal vibration in 3 μ-16μ region. The diagnostic IR bands are shown in TABLE 3.

CONCLUSION

The mixed triazole, benzimidazole derivative show positive antibacterial properties as well as antifungal effect. The antifungal properties of retrosubstituted products are larger than other.

ACKNOWLEDGEMENT

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REFERENCE

- [1] N.S.Habib, S.A.El Hawash; *IIFarmaco*, **44**, 1225-1232 (1989).
- [2] V.D.Anton, C.Wai-Keung, V.D.Anna, C.Lai-Wah, *J Fluorine chem.*, **126**, 759 (2005).
- [3] T.C.Kuhler, M.Swanson, B.Christenson, A.Klintonberg, B.Lamm, J.Fagerhag, M.Olwegard-Halvarson, V.Shecherbuehin, T.Elbrring, J.Sjstrom; *J Med Chem*, **45**, 4282 (2002).
- [4] M.T.Omar; *Egypt J PharmaSci*, **38**, 281 (1997).
- [5] K.T.Hopkins, W.D.Wilson, B.C.Bender, D.R.McCurdy, D.W.Boykin, *J Med Chem*, **41**, 3872 (1998).
- [6] G.Hakan, O.Scrif, S.Yildez, D.W.Boykin; *Eur J Med Chem*, **40**, 1069 (2005).
- [7] I.Yildiz-Oren, I.Yalcin, E.Aki-Sener, N.Ukarturk; *Eur J Med Chem*, **39**, 291 (2004).
- [8] N.S.Pawar, D.S.Dalal, S.R.Shimpi, P.P.Mahulikar; *Eur J Phar Sci*, **21**, 115 (2004).
- [9] M.A.Ismail, R.Brun, F.A.Tanious, T.Wenzler, W.D.Wilson, D.W.Boykin; *Bioorg Med Chem*, **12**, 5405 (2004).
- [10] A.M.Ismail, M.Y.Yousif, M.A.Metwally, M.M.El-Kerdawy, *Indian J Chem*, **23B**, 489 (1984).
- [11] H.Goker, C.Kus, U.Abbasoglu; *Arch. Pharm. (Weinheim)*, **328**, 425-430 (1995).
- [12] L.Garuti, M.Roberti, G.Gentilomi; *IIFarmaco*, **55**, 35-39 (2000).
- [13] K.Shinichi, F.Kosaku, F.Takashi; *Chem. Abstr. PCT. Int. Appl. WO.0105402.*, **134**, 131531g (2001).
- [14] F.Janssens, J.Torremans, M.Janssen, R.A.Stokbroekx; *J. Med. Chem.*, **28**, 1925-1933 (1985).
- [15] S.Uzunoglu, A.Tosun, T.Ozden, E.Yesilada, Hacettepe; *Univ.Eczacilik. Fak. Derg.(Persian)*, **17**, 17-24 (1997).
- [16] Q.Fanqi, D.Xinrong, L.Xinxiang, L.Xin, H.Xiaoling; *J. Nat.Sci.*, **3**, 201-204 (1998).
- [17] U.Sermin, C.Bilge, A.Tosun, S.M.Fethi, O.Berrin, Hacettepe; *Univ.Eczacilik. Fak. Derg.(Persian)*, **17**, 47-53 (1997).
- [18] H.N.Samuel, S.M.Rida, E.A.M.Badawey, H.T.Y.Fahmy, H.A.Ghozlan; *Pharmazie*, **52**, 346-350 (1997).
- [19] G.Laura, R.Marinella, P.Annalisa, L.Emanuela; *Bioorg. Med Chem.Lett.*, **11**, 3147-3149 (2001).
- [20] AnshulChawla, RamandeepKaur and AnjuGoyl.*J.Chem.Pharma Res*, **3(6)**, 925-944 (2011).
- [21] L.C.R.Carvalho, M.M.B.Marques, E.Fernandes; *Chem.Eur.J.*, **17**, 12544-12555 (2011).
- [22] (a) H.M.Eisa, A.M.Barghash, S.M.Badr, A.A.Farahat; *Indian J.Chem.*, **49B**, 1515-1525 (2012).; (b) D.K.Ravishankara, P.G.Chandrashekara; *Eur J.Chem.*, **3(3)**, 359-362 (2012).
- [23] K.Nakamoto; *The Infrared and Raman Spectra of Inorganic and Coordination Compound*. Wiley Interscience, New York, (1978).
- [24] L.J.Belamy; *The Infrared Spectra of Complex Molecule*. 2nd Edition, Chapman and Hall, London, (1980).
- [25] M.D.Whitehead; *Phytopath.*, 550 (1952).
- [26] R.K.Dubey, A.Mariya; *J.IndianChem.Soc.*, **89**, 51 (2012).
- [27] (a) C.H.Collins; *Microbiological methods*, (Butterworths, London), 92 (1964).; (b) V.Lorian; *Antibiotics in Laboratory Medicine* (Williams and Wikins, Baltimore, London), 1014 (1980).