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# STUDY OF PHYSICO-CHEMICAL PROPERTIES OF DRUG AND PHYSIOLOGICAL VARIATION IN LEAVES OF ANDROGRAPHIS PANICULATA (BURM. F.) NEES

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## ABSTRACT

Samples of Andrographis paniculata (Burm. f.) Nees of different age i.e. after 30, 60, 90, 110 and 130 days of plantation with samples of different location were studied. Results of extractive value determination show that alcohol soluble extractive value was maximum and Pet. ether soluble extractive value was least. Among all samples studied alcohol soluble extractive value of  $L_L$  sample was highest (23.42%) followed by  $Q_L$  sample. Results of physiological variation clearly indicate that both Chlorophyll a and Chlorophyll b were more in young plants than matured ones. Results also show that as Chlorophyll content decreases Carotenoids and Anthocyanin content increases.

Key words: Andrographis paniculata, Kalmegh, Chlorophyll, Physico-chemical, Physiological variation.

## **INTRODUCTION**

Andrographis paniculata (Burm. f.) Nees belongs to the family of Acanthaceae and commonly called as Kalmegh. It is found in wild through out of plains of India especially in Tamil Nadu, Karnataka, Maharashtra, Orissa, Uttar Pradesh and Uttarakhand. Andrographis paniculata (Burm. f.) Nees is one of the important herbs among 17,000 higher plant species occurring in India, out of which more than 1000 species are used over several centuries in the traditional systems of medicine viz. Ayurveda, Siddha and Unani. According to Pushpangadan et al.<sup>1</sup>, the villagers and tribal folks spread across the length and breadth of the country make use of more than 7000 plant species through oral traditions. Nearly 3/4 of the herbal drugs and perfumery products used in the world are available naturally in India. Therefore, the rich and varied plant diversity, especially the genetic diversity of medicinal and aromatic plants, is one of India's important strengths. In a significant study from China showed that not only plants growing in different geographical areas with different morphological characteristics could have different chemical constituents and physiological variation but also plants with similar morphological features and growing on the same site may have different contents of chemical constituents and physiological variation<sup>2</sup>. Andrographis paniculata or Kalmegh is one of the most widely used plants in ayurvedic formulations<sup>3</sup>. Study of physico-chemical properties of the drug, change in chlorophyll, carotenoids and anthocyanins amount with the increasing days of plantation of plant is very important because it directly relates with effectiveness and the quality of medicinal plant.

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Two types of chlorophyll are found in plants - chlorophyll a and chlorophyll b. Both chlorophylls absorb light most strongly in the red and violet parts of the spectrum. Chlorophyll is responsible for photosynthesis i.e. for the food of plant hence for the chemical constituents found in plants, which directly affects the medicinal values of plant.

Carotenoids are organic pigments that are found in the chloroplasts and chromoplasts of plants. Carotenoids in general absorb blue light. They serve two key roles in plants: they absorb light energy for use in photosynthesis, and they protect chlorophyll from photodamage<sup>4</sup>. Carotenoids have many physiological functions. Given their structure, carotenoids are efficient free-radical scavengers, and they enhance the vertebrate immune system. There are several dozen carotenoids in foods people consume, and most carotenoids have antioxidant activity<sup>5</sup>.

Anthocyanins are water-soluble vacuolar pigments that may appear red, purple, or blue depending on the pH. They are odorless and nearly flavorless. Anthocyanins are derivatives of anthocyanidins, which include pendant sugars<sup>6</sup>. Anthocyanins can be used as pH indicators because their color changes with pH; they are pink in acidic solutions (pH < 7), purple in neutral solutions (pH ~ 7), greenish-yellow in alkaline solutions (pH > 7), and colourless in very alkaline solutions, where the pigment is completely reduced<sup>7</sup>. Anthocyanins also act as powerful antioxidants. Richly concentrated as pigments in berries, anthocyanins were the topics of research presented at a 2007 symposium on health benefits that may result from berry consumption<sup>8</sup>. Laboratory-based evidence was provided to demonstrate potential health effects against: cancer, aging and neurological diseases, inflammation, diabetes, bacterial infections and fibrocystic disease<sup>9</sup>.

Therefore, in order to assure the effectiveness and the quality of the medicinal plants, it is necessary to monitor availability of the bioactive constituents both in terms of quantitative and qualitative aspects. Hence study of location variations of a particular species is an important issue with respect to physico-chemical properties and chemical constituents present, in finding pharmaceutical potential.

## EXPERIMENTAL

#### Plant material

The whole plant material, only leaves, samples of whole plant material at different stages of life cycle and samples of different location variation were used. Samples after 30, 60, 90, 110 and 130 days of plantation were collected.

Samples of whole plant material at different stages of life cycle were-

A-30 days, B-60 days, C-90 days, D-110 days (after 110 days of plantation i.e., just before flowering), E-130 days (At maturity of the crop i.e., bearing flowers, fully matured seed capsules etc.).

Leaves and whole plant material of *Andrographis paniculata* grown in natural environment (wild) from different locations were used as samples with self cultivated *Andrographis paniculata*. Samples of different location variation were named as –

- $L_W$  Dehradun self growned whole plant material
- $L_L$  Dehradun self growned leaves
- $M_W$  FRI, -Dehradun (Uttarakhand) whole plant material
- M<sub>L</sub> FRI, -Dehradun (Uttarakhand) Leaves

- N<sub>W</sub> Patanjali Ayurved, Haridwar (Uttarakhand) whole plant material
- N<sub>L</sub> Patanjali Ayurved, Haridwar (Uttarakhand) Leaves
- Ow Selaqui Dehradun (Uttarakhand) whole plant material
- OL Selaqui Dehradun (Uttarakhand) Leaves
- $\mathbf{P}_{\mathbf{W}}$  Balawala (Uttarakhand) whole plant material
- **P**<sub>L</sub> Balawala (Uttarakhand) Leaves
- $\mathbf{Q}_{\mathbf{W}}$  Himalaya Drug (Uttarakhand) whole plant material
- **Q**<sub>L</sub> Himalaya Drug (Uttarakhand) Leaves

#### Study of the powder materials (drug)

The whole plant materials of *Andrographis paniculata* were dried under shade and were powdered using a homogenizer and whole powder was considered as drug. Various chemical tests were performed according to conventional methods given in Lab manuals of Pharmacognosy by Lala<sup>10</sup>, Wallis<sup>11</sup> and Kokate<sup>12</sup>.

## **Organolaptic properties**

Organolaptic Properties of drug were examined according to conventional methods given by Kokate<sup>12</sup>. The green powder was treated with some acidic, basic and neutral routinely used reagents and characteristic changes were observed. Results are given in Table 1.

#### **Fluorescence behaviors**

The fluorescence behavior of the powder of *Andrographis paniculata* as such and after treating with some chemical reagents was determined according to the methods of Chase and Pratt<sup>13</sup>. The dried powder sample was treated with various chemical reagents like  $H_2SO_4$ , HCl, HNO<sub>3</sub> etc. The mixture was mixed well, allowed to stand for few minutes and filtered. The filtrate was examined under both U.V. and visible light. Fluorescence characteristics of the powder were observed in day light as well as in ultraviolet radiation at 254 nm and 366 nm. The solvents used for the extraction procedure were Acetone, chloroform, ethanol, hexane, methanol, petroleum ether and water. After extraction all extracts were observed in day light as well as in ultraviolet (UV) light at 254 nm and 366 nm and colour was noted. Results are given in Table 2 and 3.

#### Physico-chemical properties of the drug

Physico-chemical constants like ash value, water soluble extracts, alcoholic extracts, loss on drying and pH values were determined as per method described in Indian Pharmacopoeia<sup>14</sup>.

#### Melting point of the drug

The melting point of the drug was determined by melting point apparatus. The shade dried coarse powder of the leaves and whole plant material of *Andrographis paniculata* was examined and it matches with the value as given in Indian Pharmacopoeia.

### pH values

The shade dried coarse powder of the leaves and whole plant material of *Andrographis paniculata* was examined. pH value of drug was determined by pH meter (QC/Micro/pH 01Sr No. 391505).

Results are given in Table 4.

#### **Determinations of ash values**

Ash is the inorganic residue left after ignition at 650-700°C. The ash content is an approximate measure of the mineral content and other inorganic matter in biomass. The ash content is used in conjunction with other assays to determine the total composition of biomass samples. Methods used were according to Harborne<sup>15</sup>.

#### Apparatus

Silica crucibles

Furnace - An electric furnace was used for igniting the samples. Furnace was fitted with an indicating pyrometer, so that the desired temperature can be maintained.

Analytical balance - sensitive to 0.1 mg.

Desiccator, Drying oven - with temperature control of  $105 \pm 2^{\circ}$ C.

(1) Total ash (TA) value: This value was determined using a minimum of 2.0-3.0 g of material in a furnace heated gradually to the ignition temperature of 650-700°C. Accurately 2 to 3 g of air-dried samples of *Andrographis paniculata* were weighed in a tared silica dish and incinerate at a temperature not exceeding 700°C until ash free from carbon is obtained. Then it was cooled and weighed. The process was repeated until at least two consecutive constant weights were obtained. The results are expressed as range or mean value  $\pm$  standard deviation. The percentage of ash was calculated with reference to the air – dried drug.

(2) Determination of acid insoluble ash (AIA): Above obtained ash was boiled with 25 mL of 2 M hydrochloric acid for 5 min, the insoluble matter was collected in a Gooch crucible or on an ash less filter paper, washed with hot water, ignited and cooled in desiccators. Then it was weighed. The percentage of acid insoluble ash was calculated with reference to the air-dried drug.

(3) Determination of water soluble ash (WSA): Ash was boiled for 5 min. with 25 mL of water; the insoluble matter was collected in a Gooch crucible or on an ash less filter paper, washed with hot water, and ignited for 15 min at a temperature not exceeding 450°C. The weight of insoluble matter was subtracted from the weight of the ash, the difference in weight represent the water-soluble ash. The percentage of water-soluble ash was calculated with reference to the air-dried drug.

Formula used for the calculation of percentage of all three types of Ash Values is -

Ash % = 
$$\frac{\text{Loss in weight}}{W} \ge 100$$

W = Weight of air - dried drug. Results are given in Table 5.

#### Loss on drying (LOD)

Methods used were according to Brain<sup>16</sup>. Loss on drying of the air-dried samples of *Andrographis paniculata was* analyzed. This was carried out using a minimum of 0.5-1.0 g of material. Accurately weighed quantity of sample was taken in a tared glass bottle and initial weight was taken. The sample was heated in a Lindberg/Blue M gravity-convention oven maintained at 105-110°C, for 3 h, after which the sample was allowed to cool to room temperature in desiccators, and subsequently weighed. The time interval

from the oven to point of weighing was usually about 30 minutes. This procedure was repeated until a constant weight was obtained.

Loss on drying (%) = 
$$\frac{\text{Loss in weight}}{W} \times 100$$

Where W = weight of the sample powder in gms.

The results are expressed as a range or as mean  $\pm$  standard deviation. Results are given in Table 6.

### **Determination of extractive values**

Methods used were according to Kokate et al.<sup>12</sup> The extractive values for various solvents of airdried sample were evaluated.

- (i) Water-soluble extractives
- (ii) Alcohol soluble extractives
- (iii) Pet ether soluble extractives

(1) Evaluation of water extractive value: About 5 g of accurately weighed coarsely powdered, airdried sample was transferred into a glass-stoppered, 250 mL reflux conical flask, followed by the addition of 50 mL of boiled water. The flask was well shaken, and allowed to stand for 10 minutes. It was cooled and filtered. Filtrate was transferred to an evaporating dish, which was 7.5 cm in diameter; the solvent was evaporated on water bath, allowed to dry for 30 minutes, finally dried in an oven and residue was weighed. Percentage of water-soluble extractives was calculated with reference to the air-dried drug.

(2) Alcohol soluble extractive value: 5 g of dried samples of *Andrographis paniculata* were macerated with 100 mL of alcohol in a closed flask, shaking frequently during the first 6 hours and allowed to stand for 18 hours separately. Thereafter, it was filtered rapidly taking precaution to minimize the loss of methanol. Evaporated 25 mL of filtrate to dryness in a tared flat bottom shallow dish dried at 105°C and weighed. Percentage of Alcohol soluble extractive was calculated with reference to the air-dried samples.

(3) Pet. ether soluble extractive value: 5 g of dried samples of *Andrographis paniculata* were macerated with 100 mL of pet ether (40-60°C) in a conical flask, shaking frequently during the first 6 hours and allowed to stand for 18 hours separately. Thereafter, it was filtered rapidly taking precaution against loss of pet ether. Evaporated 25 mL of filtrate to dryness in a tared flat bottom shallow dish dried at 105°C and weighed. Percentage of pet ether soluble extractive was calculated with reference to the air-dried samples.

Results are given in Table 7.

## **Physiological variation**

The experiment was conducted for the young and mature plants of selected populations of *Andrographis paniculata (Burm. f.) Nees.* Young (60 days old) and mature (130 days old) plants were selected for analyzing age-related changes in chlorophyll, carotenoid and anthocyanin content among different samples. Fresh leaves of different samples of *Andrographis paniculata* were used.

(1). Variation in Chlorophyll and Carotenoids: Chlorophyll and carotenoids were extracted using 80% acetone as shown by Sadasivam and Manikkam<sup>17</sup> and also by Sabu<sup>18</sup>. 5 g fresh leaves were

homogenized in 40 mL of 80% acetone and then the extract was centrifuged at 1000 rpm for 2 min in Remi (S/N0. BCBR-2182) cooling centrifuge. The supernatant was collected and used for measuring optical density at 490, 645 and 663 nm using UV10 Thermo scientific visible spectrophotometer (S/No. HEDM218008).

(2). Variations in Anthocyanin content: For estimating variation in anthocyanin content of leaves, 0.5 g leaf tissue was homogenized in 4 mL of methanol containing 1.0 M HCl and was kept at 4°C for 4 h. The homogenate was filtered and centrifugedat 10,000 rpm for 30 min and the absorbance of the supernatant was measured at 530 nm<sup>19</sup>, after suitable dilution.

Results are given in Table 8.



Green leaves of young plant.



Coloured leaves of matured plant.

## **RESULTS AND DISCUSSION**

Organolaptic Properties: Coarse Powder

Colour-Green Foreign matter- < 1.0% Sand & Silica- Absent Odour- Slight characteristic Taste- Intensely bitter Insect infestation- Nil Rodent contamination- Nil Powder pressed between two filter papers for 24 hours-Stained

S. No.	Chemical treatment	Observation
1	Drug Powder treated with HCI	Powder settles down slowly Colour: Light brown
2	Drug green Powder treated with Conc. HNO <sub>3</sub>	Powder float on the surface. Colour: Yellowish brown
3	Drug green Powder treated with Conc. $H_2SO_4$	Powder settles down slowly. Colour: Brownish black
4	Drug green Powder treated with 5% aqueous NaOH	Powder settles down slowly Colour: Brownish
5	Drug green Powder treated with iodine solution	Powder settles down immediately. Colour: Brownish
6	Drug green Powder treated with 5% aqueous KOH	Powder settles down slowly. Colour: Dark brown
7	Drug green Powder treated with Glacial Acetic Acid	Powder settles down immediately. Colour: Light Brown
8	Drug green Powder treated with 5% aqueous FeCI <sub>3</sub>	Powder settles down immediately. Colour: Greenish

Table 1: Behavior of Andrographis paniculata (drug) with different reagents

## Fluorescence behavior of Andrographis paniculata (drug powder) without any extraction

Drug powder of *Andrographis paniculata* (whole plant material) without any extraction was observed in day light as well as in ultraviolet (UV) light at 254 nm and 366 nm and colour was noted.

 Table 2: Fluorescence behavior of Andrographis paniculata (drug powder)

S. No.	Chemical treatment	Under ordinary light	Under UV light 254 nm	Under UV light 366 nm	
1	Drug green powder as such Green		Green	Silver green	
2	Drug green powder treated with 1N NaOH in methanol	C		Cream	
3	Drug green powder treated with 1 N NaOH in water			Dark greenish white	
4	Drug green powder treated with 50% HNO <sub>3</sub>	5		Greenish yellow	
5	Drug green powder treated with $50\%$ H <sub>2</sub> S0 <sub>4</sub>	Yellowish brown	Yellowish green	Greenish white	

## Fluorescence behavior of the drug powder after extraction with chemical reagents

Whole plant material of *Andrographis paniculata* was extracted. Then extracts were observed in day light as well as in ultraviolet (UV) light at 254 nm and 366 nm and colour was noted.

S. No.	Extractives	Under ordinary light	Under UV light (254 nm)	Under UV light (366 nm)	
1	Hexane	Light green	Colourless	Colourless	
2	Pet. ether	Yellowish	Light green	Colourless	
3	Chloroform	Yellowish	Light green	Colourless	
4	Acetone	Green	Light green	Umber	
5	Ethyl alcohol	Green	Dark green	Umber	
6	Methyl alcohol	Green	Dark green	Umber	
7	Dist. water	Green	Pale green	Umber	

Table 3: Fluorescence behavior of Andrographis paniculata extracts

**Physico-chemical properties of the drug: 1. Melting point of the drug:** The melting point of the drug, determined by melting point apparatus. Average value of triplicates is given in the table. It is between 226-229°C, which matches with the standard melting point value (229°C) of drug as given in Indian Pharmacopoeia.

**pH values of the drug:** pH value of shade dried coarse powder of the leaves and whole plant material of *Andrographis paniculata* was examined. pH of 1% aqueous solution and pH of 10% aqueous solution was measured. Average value of triplicates is given in the table.

Table 4: Melting point and pH values of Different samples of Andrographis paniculata

S. No.	Samples of Andrographis paniculata	Melting point	pH values 1% aq. sol.	pH values 10% aq.sol.	
1	Whole plant material	228.5°C	7.94	8.22	
2	Leaves	226°C	7.82	7.64	

## **Determination of ash values**

Three ash values were calculated. They are-

1) Total Ash (TA) Value

#### 2) Determination of Acid Insoluble Ash (AIA)

## 3) Determination of Water Soluble Ash (WSA)

Samples of whole plant material at different stages of life cycle and Samples of different location variation were used. The results are expressed as range or mean value  $\pm$  standard deviation. The percentage of ash was calculated with reference to the air – dried drug. Result is reported up to two decimal places, as a percentage of the samples.

Table 5: Ash values of different samples of Andrographis paniculata.

S. No.	Samples of Andrographis paniculata	Total ash (% w/w)	Acid insoluble Ash (% w/w)	Water soluble Ash (% w/w)
1	Α	10.12%	1.22%	1.13%
2	В	10.14%	1.24%	1.14%

S. No.	Samples of Andrographis paniculata	Total ash (% w/w)	Acid insoluble Ash (% w/w)	Water soluble Ash (% w/w)
3	С	9.84%	1.18%	1.23%
4	D	9.24%	0.98%	1.02%
5	Ε	12.82%	1.97%	1.42%
6	$\mathbf{L}_{\mathbf{W}}$	11.79%	1.62%	1.65%.
7	$L_{L}$	10.02%	1.11%	1.16%
8	$\mathbf{M}_{\mathbf{W}}$	11.84%	1.99%	1.74%
9	$M_L$	11.01%	1.84%	1.66%
10	$N_{W}$	11.65%	2.03%	1.53%
11	$N_L$	11.78%	1.74%	1.25%
12	$\mathbf{O}_{\mathbf{W}}$	12.65%	1.64%	1.51%
13	$O_L$	11.08%	1.82%	1.08%
14	$\mathbf{P}_{\mathbf{W}}$	10.52%	1.97%	1.21%
15	$\mathbf{P}_{\mathrm{L}}$	11.24%	2.01%	1.60%
16	$\mathbf{Q}_{\mathbf{W}}$	12.14%	1.88%	1.32%
17	$Q_{L}$	11.99%	1.74%	1.66%

## Loss on drying (LOD)

Loss on drying of the air-dried samples of. *Andrographis paniculata* was carried out using Lindberg/Blue M gravity-convention oven maintained at 105-110°C. The results are expressed as a range or as mean  $\pm$  standard deviation.

S. No.	Samples of Andrographis paniculata	Loss on drying (LOD) (% $\rm w/\rm w)$
1	Α	8.40%
2	В	8.62%
3	С	9.45%
4	D	9.12%
5	E	10.45%
6	$\mathbf{L}_{\mathbf{W}}$	10.20%
7	$\mathbf{L}_{\mathbf{L}}$	10.24%
8	$\mathbf{M}_{\mathbf{W}}$	11.22%
9	$\mathbf{M}_{\mathbf{L}}$	11.26%
10	$\mathbf{N}_{\mathbf{W}}$	10.98%
11	$\mathbf{N}_{\mathbf{L}}$	11.02%
12	$\mathbf{O}_{\mathbf{W}}$	11.43%
13	$O_L$	11.08%
14	$\mathbf{P}_{\mathbf{W}}$	9.99%
15	$\mathbf{P}_{\mathbf{L}}$	10.03%

Table 6: Loss on drying (LOD) (% w/w) of Different samples of Andrographis paniculata

Deterioration time of the plant material depends upon the amount of water present in plant material. If the water content is high, the plant can be easily deteriorated due to fungal attack. To determine chemical constituents correctly it is necessary to calculate LOD.

## **Determination of extractive values**

Three extractive values were calculated. They are-

- (1) Evaluation of water extractive value
- (2) Alcohol soluble extractive value
- (3) Pet. ether soluble extractive value

#### Table 7: Extractive values of different samples of Andrographis paniculata

S. No.	Samples of Andrographis paniculata	Water extractive value	Alcohol soluble extractive value	Pet. Ether soluble extractive value
1	Α	12.24%	18.45%	10.96%
2	В	12.25%	18.99%	11.23%
3	С	14.82%	19.82%	11.16%
4	D	16.24%	20.88%	10.88%
5	Ε	13.98%	20.42%	11.45%
6	$\mathbf{L}_{\mathbf{W}}$	16.13%	21.12%	10.99%
7	$\mathbf{L}_{\mathbf{L}}$	17.02%	23.42%	11.02%
8	$\mathbf{M}_{\mathbf{W}}$	15.98%	20.45%	11.43%
9	$\mathbf{M}_{\mathbf{L}}$	16.79%	21.23%	11.62%
10	$\mathbf{N}_{\mathbf{W}}$	14.97%	21.22%	11.03%
11	$N_L$	16.82%	22.15%	11.32%
12	$\mathbf{O}_{\mathbf{W}}$	14.85%	21.17%	11.23%
13	$O_L$	16.05%	21.65%	11.14%
14	$\mathbf{P}_{\mathbf{W}}$	16.49%	20.43%	10.89%
15	$\mathbf{P}_{\mathbf{L}}$	16.89%	19.69%	11.21%
16	$\mathbf{Q}_{\mathbf{W}}$	15.73%	21.48%	11.14%
17	$Q_{L}$	16.11%	22.88%	11.29%

#### A-30 days, B-60 days, C-90 days, D-110 days, E-130 days of plantation

- $\boldsymbol{L}_{\boldsymbol{W}}$  Dehradun self growned whole plant material
- $M_W$  FRI, -Dehradun (Uttarakhand) whole plant material
- $N_W$  Patanjali Ayurved, Haridwar (Uttarakhand) whole
  - plant material
- $O_W$  Selaqui Dehradun (Uttarakhand) whole plant material
- $\mathbf{P}_{\mathbf{W}}$  Balawala (Uttarakhand) whole plant material
- $\boldsymbol{Q}_{\boldsymbol{W}}$  . Himalaya Drug (Uttarakhand) whole plant material
- $\boldsymbol{L}_L$  -Dehradun self growned Leaves
- $M_L$  FRI, -Dehradun (Uttarakhand) Leaves
- N<sub>L</sub> Patanjali Ayurved, Haridwar (Uttarakhand) Leaves
- O<sub>L</sub> Selaqui Dehradun (Uttarakhand) Leaves
- $P_L$  Balawala (Uttarakhand) Leaves
- $Q_L$  Himalaya Drug (Uttarakhand) Leaves

Results show that alcohol soluble extractive value is maximum and Pet. ether soluble extractive value is least. Among all samples studied alcohol soluble extractive value of  $L_L$  sample is highest (23.42%) followed by  $Q_L$  sample.

#### **Physiological variation**

There are a bunch of different equations for calculating amounts of chlorophyll in plants. Arnon's<sup>20</sup> equations for calculation of chlorophyll extracted in 80% acetone were proven by Hiscox & Israelstam<sup>21</sup>. These equations are used for calculating amounts of chlorophyll in different samples of *Andrographis paniculata*.

Arnon<sup>20</sup> reported the following equations for quantification of the total chlorophyll, chlorophyll a and chlorophyll b content in an 80% acetone extract:

Total chlorophyll ( $\mu g/mL$ ) = 20.2 (A<sub>645</sub>) + 8.02 (A<sub>663</sub>)

Chlorophyll a ( $\mu$ g/mL) = 12.7 (A<sub>663</sub>) – 2.69 (A<sub>645</sub>)

Chlorophyll b ( $\mu$ g/mL) = 22.9 (A<sub>645</sub>) – 4.68 (A<sub>663</sub>)

Where  $A_{663}$  is the solution absorbance at 663 nm and  $A_{645}$  is the absorption at 645.

Optical density of different samples of *Andrographis paniculata* was measure at 490, 645 and 663nm using UV10 Thermo scientific visible spectrophotometer (S/No. HEDM218008).

 Table 8: Chlorophyll, Carotenoids and Anthocyanin content of Different samples of Andrographis paniculata

S. No.	Samples of Andrographis paniculata	Total chlorophyll mg/plant tissue	Chlorophyll a mg/plant tissue	Chlorophyll b mg/plant tissue	Carotenoids mg/plant tissue	Anthocyanin mg/plant tissue
1	YL <sub>1</sub>	0.067	0.049	0.018	3.423	0.892
2	YL <sub>2</sub>	0.055	0.042	0.013	4.121	1.213
3	$ML_1$	0.048	0.036	0.012	5.627	1.567
4	$ML_2$	0.045	0.034	0.011	4.891	1.712
$YL_1$ , $YL_2 = Young$ (60 days old) Leaves, $ML_1$ , $ML_2$ =mature (130 days old) Leaves of Andrographis paniculata						

Results clearly indicate that both Chlorophyll a and Chlorophyll b are more in young plants than matured ones. Results also show that as Chlorophyll content decreases, Carotenoids and Anthocyanin content increases.

## CONCLUSION

Study of physico-chemical properties of the drug of different locations and study of physiological variation at different stages of life cycle of *Andrographis paniculata* will serve as the pioneering project to assess this medicinally important native plant and future research on other medicinal native plants of Uttarakhand based on the fact that plants from different sources may have different phytochemical profiles and pharmacological actions.

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