

SIMULTANEOUS SPECTROPHOTOMETRIC ESTIMATION OF CILOSTAZOL AND ASPIRIN IN SYNTHETIC MIXTURE

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ABSTRACT

Two simple, rapid, accurate and economical methods have been developed for the simultaneous estimation of aspirin and cilostazol in the synthetic mixture. Aspirin has absorbance maxima at 226 nm and cilostazol has absorbance maxima at 257 nm in methanol. The linearity was observed in the concentration range of 2-24 µg/mL for the both aspirin and cilostazol. First method is based on the simultaneous equations; Absorbances of both the drugs were determined at 226 nm (λ_{max} of aspirin) and at 257 nm (λ_{max} of cilostazol). The method was validated in terms of accuracy (99.30 ± 0.97, 101.58 ± 0.89). Second method is based on Q-absorbance ratio; absorbances of both the drugs were determined at 226 nm (λ_{max} of aspirin) and at isoabsorptive point (239.5 nm). Q-absorption ratio method was validated in terms of accuracy (98.05 ± 1.20, 99.12 ± 1.00). The proposed methods were found accurate, reproducible and economical for the routine analysis of both the drugs in the synthetic mixture.

Key words: Spectrophotometric, Cilostazol, Aspirin

INTRODUCTION

Cilostazol is chemically 6-[4-l(-cyclohexyl-lH-tetrazol-5-yl-butoxyl] 3-4-dihydro-2(1H) - quinolinone¹. Cilostazol is the member of the phosphodiesterase inhibitor-3, approved by the US Food and Drug Administration (FDA) in 1999. It has been proved effective in significantly improving walking distances among patients with claudication²⁻⁴. Aspirin is chemically 2-(acetyloxy) benzoic acid; Acetyl salicylic acid; salicylic acid acetate⁵. Aspirin is official in IP⁶, BP⁷, USP⁸ and in the extra pharmacopoeia⁹.

A survey of literature revealed HPLC^{10. 11} and LC/MS/MS¹² method for the determination of cilostazol in biological fluids and tablets. Cilostazol and aspirin are

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co-administred in patients with co-existent 1C and coronary artery disease^{13, 14} and therefore, in future combination of these two drugs comes in market.

In the present investigation an, attempt has been made to develop two simple, economical, accurate and reproducible spectrophotometric methods for the simultaneous estimation of cilostazol and aspirin in the synthetic mixture. First method is based on simultaneous equation method and second method is based on Q-absorbance ratio. The proposed methods were successfully applied for simultaneous estimation of cilostazol and aspirin in the synthetic mixture.

EXPERIMENTAL

Materials and methods

Shimadzu model 1601 double beam UV-visible spectrophotometer with a pair of 10 mm matched quartz cells was used to measure absorbance of the resulting solutions. Sartorius CP224S analytical balance, an ultrasonicater (Frontline FS 4). cilostazol (Cadila Pharma Ltd., Ahmedabad), aspirin and methanol, AR grade (Merck India limited, Mumbai) were used in the study.

Preparation of standard solutions and synthetic mixture

Standard cilostazol stock solution (100 μ g/mL): Accurately weighed cilostazol (10 mg) was transferred in 100 mL volumetric flask, dissolved in methanol and diluted to the mark with methanol.

Standard aspirin stock solution (100 μ g/mL): Accurately weighed Aspirin (10 mg) was transferred in 100 mL volumetric flask, dissolved in absolute alcohol and diluted to the mark with absolute alcohol.

Preparation of synthetic mixture of cilostazol and aspirin

The synthetic mixture of cilostazol and aspirin was prepared in the ratio of 1 : 0.7. Accurately weighed cilostazol (10 mg) and aspirin (7.5 mg) were transferred in 100 mL volumetric flask and dissolved in methanol (70 mL). Common excipients, which are used in the tablet formulation, were added in this mixture and sonicated for 20 minutes. This solution was filtered through the Whatman filter paper No. 41 and the residue was washed thoroughly with methanol. The filtrate and washings were combined and diluted to the mark with methanol to get solution having cilostazol (100 μ g/mL) and aspirin (75 μ g/mL).

Selection of wavelength for estimation of cilostazol and aspirin

The standard stock solutions of cilostazol and aspirin were scanned in the range of 200 nm to 400 nm against methanol as a blank. Maximum absorbance was obtained at 257 nm and 226 nm for cilostazol and aspirin, respectively. Isoabsorptive point was found at 239.5 nm (Fig. 1).

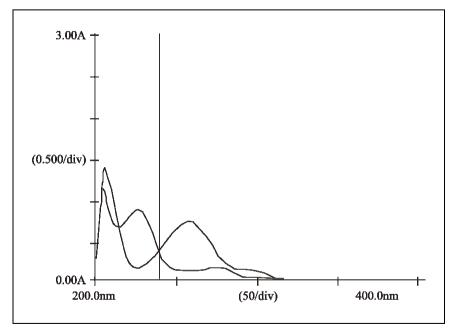


Fig. 1: Overlain spectra of cilostazol and aspirin showing isoabsorptive point at 239.5 nm

Calibration curve for cilostazol and aspirin

A calibration curve was plotted over a concentration range 4-24 μ g/mL for both cilostazol and aspirin. Accurately measured standard stock solution of cilostazol (0.4, 0.6, 0.8, 1.0, 1.2, 1.4, 1.6, 1.8, 2.0 and 2.4 mL) and standard stock solution of aspirin (0.4, 0.6, 0.8, 1.0, 1.2, 1.4, 1.6, 1.8, 2.0 and 2.4 mL) were transferred into a series of 10 mL of volumetric flasks and diluted to the mark with methanol. Absorbance of each solution was measured at both the wavelength 257 nm and 226 nm. Calibration curves were constructed for cilostazol and aspirin by plotting absorbance versus concentrations at both wavelengths. Each reading was average of three determinations. In Q- absorbance ratio method, absorbance of each solution was measured at the three wavelengths 226 nm, 237 nm and 239.5 nm. Calibration curves were constructed for cilostazol and aspirin by were were constructed for cilostazol and aspirin was measured at the three wavelengths 226 nm, 237 nm and 239.5 nm. Calibration curves were constructed for cilostazol and aspirin by blotting was measured at the three wavelengths 226 nm, 237 nm and 239.5 nm.

plotting absorbance versus concentrations at three wavelengths. Each reading was average of three determinations.

Accuracy

Accuracy was determined in term of percent recovery. The proposed method was applied to determine cilostazol and aspirin in the synthetic mixture. The recovery experiments were carried out in triplicate by spiking previously analyzed samples of the synthetic mixture with three different concentrations of standards.

Analysis of the synthetic mixture

The absorbance of final sample solution was measured against methanol as a blank at 226, 239.5 and 257 nm. The amount of cilostazol and aspirin were calculated using simultaneous equations as well as Q-absorbance ratio method.

	257 nm		226 nm		239.5 nm
Parameters	Aspirin	Cilostazol	Aspirin	Cilostazol	Cilostazol and aspirin
Beer's law limit (µg/mL)	4-24	4-24	4-24	4-24	4-24
Molar absorptivity (lit/mole/cm)	$0.050 imes 10^4$	1.768×10^4	0.837×10^4	0.321×10^{4}	$\begin{array}{c} 0.512 \times 10^4 \\ (\text{Asprin}) \end{array}$
					1.049×10^4 (Cilostazol)
Sandel's sensitivity (µg/mL/cm ² /0.001)	0.3571	0.0208	0.0214	0.1149	0.03521
LOD (µg/mL)	0.378	0.486	1.206	1.359	0.899
LOQ (µg/mL)	1.126	2.181	3.928	2.915	2.548
Regression equation $(y = mx + c)$	y = 634.99 x = -0.0064	17557 +0.003	9206.2 0.0278	3303.1 +0.0003	0.035 -0.0594
Correlation coefficient (r ²)	0.9954	0.9990	0.9994	0.9949	0.9925

RESULTS AND DISCUSSION

Table 1. Summary of validation parameters for simultaneous equation and Q-absorption ratio methods

Calibration curves for cilostazol and aspirin over concentration range of 4-24 μ g/mL were plotted and molar absorptivity for both the compounds were calculated at three wavelengths 226 nm (λ_{max} of aspirin), 239.5 nm (isoabsorptive point) and 257 nm (λ_{max} of cilostazol). The linearity of the calibration graphs was validated by the high value of correlation coefficients of the regression (Table 1). LOD for cilostazol and aspirin were found to be 0.58 μ g/mL and 0.378 μ g/mL respectively while LOQ for cilostazol and aspirin were found to be 2.181 μ g/mL and 2.08 μ g/mL, respectively by both the methods. These data show that both the methods are sensitive in the determination of cilostazol and aspirin.

Accuracy

The percent recoveries obtained were 99.30 to 101.58 and 100.08 to 101.14 for cilostazol and aspirin, respectively by simultaneous equation method; 98.05 to 99.05 and 98.12 to 98.59 for cilostazol and aspirin, respectively by Q-absorbance ratio method (Table 2). The low value of SD indicates that both the methods are accurate.

Content	Amount taken (μg/mL)	Amount added (μg/mL)	% Recovery ± SD n = 5 by simultaneous equation method	% Recovery ± SD n = 5 by Q - absorption ratio method
Cilostazol	5	5	101.00 ± 1.14	98.05 ± 1.20
	5	10	$99.30\pm\!\!0.97$	98.57 ± 1.05
	5	15	101.58 ± 0.89	99.05 ± 0.90
Asprin	5	5	100.12 ± 0.99	98.59 ± 1.18
	5	10	100.08 ± 1.18	99.12 ± 1.00
	5	15	101.4 ± 1.21	98.14 ± 1.02

Analysis of the synthetic mixture

 Table 2. Data of recovery studies by simultaneous equation and Q-absorption ratio

 methods

In the simultaneous equation method-concentration of cilostazol and aspirin in the synthetic mixture were found out by solving following equations.

$$C_{c} = \frac{A_{2}a_{a1} - A_{1}a_{a2}}{A_{c2}a_{a1} - a_{m1}a_{r2}} \text{ and } C_{c} = \frac{A_{1}a_{c2} - A_{2}a_{c1}}{a_{c2}a_{a1} - a_{c1}a_{a2}}$$

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Where; C_c, C_a = concentration of cilostazol and aspirin in the sample solution
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 A_1 , A_2 = Absorbances of the sample solution at 226 nm and 257 nm, respectively

 A_{c1} and a_{c2} = Molar absorptivities of cilostazol at 226 nm and 257 nm, respectively

 A_{a1} and a_{a2} = Molar absorptivities of aspirin at 226 nm and 257 nm, respectively

In the Q- absorbance ratio method, concentration of cilostazol and aspirin in the sample solutions were calculated using equations $C_{c2}=(Q_o-Q_a/Q_c-Q_a) \times A_3/a_{c3}$ and $C_{p2}=A_3/a_{a3}-C_{c2}$, where A₁ and A₃ are absorbances of sample solution at 257 nm and 239.5 nm; and a_{c3} and a_{a3} are molar absorptivity of cilostazol and aspirin at 239.5 nm; aci and aai are molar absorptivity of cilostazol and aspirin at 257 nm. $Q_o=A_1/A_3$, $Q_c=a_{c1}/a_{c3}$ and $Q_a=a_{a1}/a_{a3}$.

The proposed validated methods were successfully applied to determine cilostazol and aspirin in the synthetic mixture. The % recoveries for cilostazol and aspirin obtained were 101.58 ± 0.89 , 101.14 ± 1.21 by simultaneous equations method and 98.57 ± 1.05 , 98.59 ± 1.18 by Q-absorption ratio method, respectively (Table 3). No interference of the excipients with the absorbance appeared; hence, the proposed methods were applicable for the quantitative determination of cilostazol and aspirin in synthetic mixture. The proposed methods are based on simultaneous equation and absorption ratio and only require measurement of absorbance at selected wavelengths. The proposed methods were found to be simple, rapid, economical, accurate and precise. They are particularly useful for routine in-process quality control and simultaneous quantification of cilostazol and aspirin in combined synthetic mixture.

Content	% Amount found ± SD n = 5 by Simultaneous equation method	% Amount found ± SD n = 5 by Q-absorption ratio method	
Cilostazol	100.48 ± 1.20	99.01 ± 1.58	
Aspirin	99.39± 1.53	101.15 ± 1.63	

 Table 3. Analysis of the synthetic mixture by simultaneous equation and

 Q-absorption ratio methods

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