



NICKEL ACETATE AS EFFICIENT ORGANOMETALLIC CATALYST FOR SYNTHESIS OF BIS (INDOLYL) METHANES

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ABSTRACT

Bis(indolyl) methanes derivatives have been synthesized using a catalytic amount of organometallic anhydrous nickel acetate at room temperature with excellent yields. The remarkable selectivity under mild, neutral and, inexpensive catalyst are attractive features.

Key words: Bis(indolyl) methanes, Nickel acetate, Aldehydes, Ketones.

INTRODUCTION

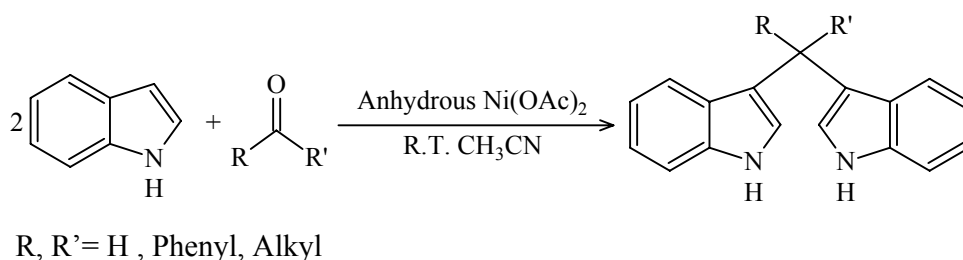
The development of simple, efficient and economically viable chemical process or methodologies for widely used organic compounds are in great demand¹. Various methods have been developed for their synthesis using Lewis acid catalysts²⁻⁸, ionic liquids⁹, trichloro-1,3,5-triazine¹⁰, and potassium hydrogen sulphate¹¹. However, many of these reported methods suffer from one or more disadvantages such as harsh experimental procedure and reagents that are expensive, moisture sensitive. A mild and efficient catalyst for the synthesis of bis(indolyl) methanes is highly desirable.

EXPERIMENTAL

In this communication, we report a synthesis of Bis(indolyl) methanes by using organometallic anhydrous nickel acetate as catalyst. A wide variety of compounds that were

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applied to the optimal reaction conditions to prepare a wide range of bis(indolyl) methanes (**Scheme 1**).



Scheme 1

General experimental procedure for Bis(indolyl) methanes

A mixture of benzaldehyde (2 mmol), Indole (4 mmol) and anhydrous Ni(OAc)₂ (0.1 mmol, 30 mg) was stirred magnetically at room temperature, acetonitrile (1 mL) and the progress of the reaction was monitored by thin-layer chromatography. The product was dried over anhydrous Na₂SO₄ and further purification by column chromatography.

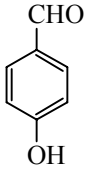
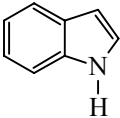
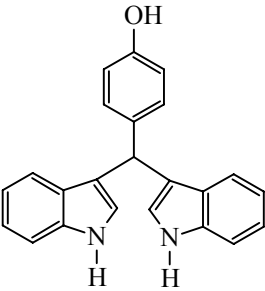
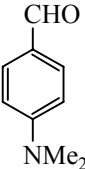
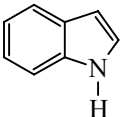
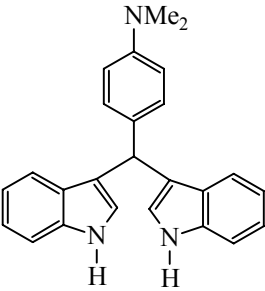
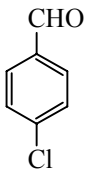
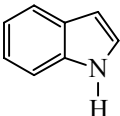
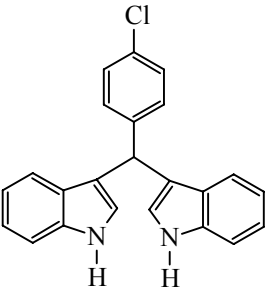
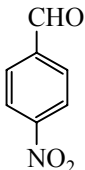
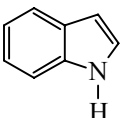
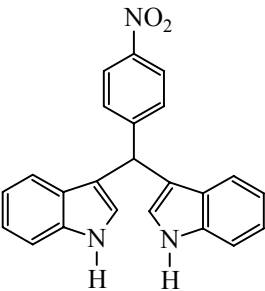
RESULTS AND DISCUSSION

The reaction proceeded efficiently and smoothly at room temperature in presence of anhydrous Ni(OAc)₂ as a catalyst, and the products were obtained in excellent yields. Various aromatic aldehydes, aliphatic aldehyde and ketones gives the corresponding products with excellent yield (Table 1, entries 1-9).

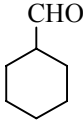
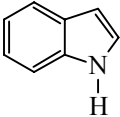
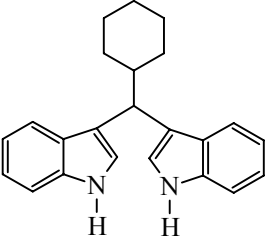
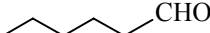
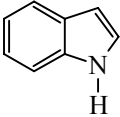
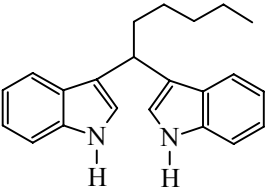
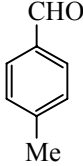
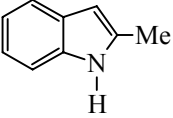
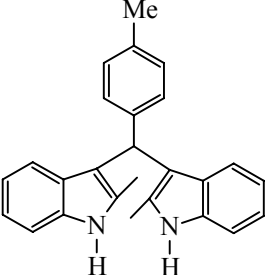
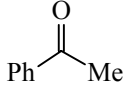
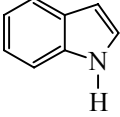
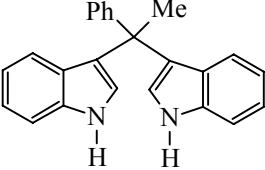
Table 1: Synthesis of bis(indolyl) methanes

Entry	Aldehyde (a)	Indoles	Product (b)	Time (min)	Yield ^c (%)
1				25	94

Cont...

Entry	Aldehyde (a)	Indoles	Product (b)	Time (min)	Yield ^c (%)
2				25	92
3				55	90
4				55	87
5				75	88

Cont...

Entry	Aldehyde (a)	Indoles	Product (b)	Time (min)	Yield ^c (%)
6				75	87
7				85	86
8				35	89
9				110	87

^aThe substrate was treated with indole (4 mmol) by stirring at room temperature with anhydrous Ni(OAc)₂ in presence of acetonitrile as solvent;
^bAll products were identified by their IR and ¹H NMR spectra;
^cIsolated yields after column chromatography

Spectral data

3,3'-Bisindolyl phenyl methane (1b): Pale-red solid, yield 94%, m.p. 122-124°C

IR (KBr): 736, 1012, 1173, 1336, 1415, 1599, 2848, 3024, 3054, 3409 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 7.8 (s, 2H); 7.1-7.4 (br m, 8H); 6.3-6.8 (m, 5H); 4.1-4.4 (s, 2NH);

2.2 (s, H): ^{13}C NMR (CDCl_3): 144.1, 136.7, 128.7, 128.6, 127.2, 126.9, 123.7, 121.9, 119.9, 111.1, 40.2. EIMS; m/z 322

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