

NEW PROPACIN ANALOGUES FROM A COLLECTION OF THE WHOLE PLANT OF *JATROPHA GOSSYPIFOLIA*¹

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ABSTRACT

From a collection of the whole plant of *Jatropha gossypifolia* two new propacin analogues, jatrocins A and B along with the known coumarino–lignans, propacin and cleomiscosin A were isolated. The structures of the new compounds were settled by extensive studies of their spectra (mainly 1D- and 2D- NMR).

Key words : New propacin analogues, *Jatropha Gossypifolia*

INTRODUCTION

Jatropha gossypifolia Linn (Euphorbiaceae), is a small shrub found in different parts of India. The plant is reputed^{2,3} ethnomedically to treat cancerous growths. Earlier chemical investigation on the plant resulted in the isolation of various diterpenes and lignans.^{4–10} In continuation of our work^{7–10} on the bioactive constituents of the plant, a collection of the whole plant has been examined. Two new propacin analogues, jatrocins A(**1**) and B(**2**) together with the known coumarino–lignans, propacin (**3**)^{11,12} and cleomiscosin A (**4**)¹³ were isolated. Here, the isolation of these constituents and the structure elucidation of the new compounds, **1** and **2** has been reported.

EXPERIMENTAL

Melting points were measured in a Buchi 510 instrument and are uncorrected. Spectra were recorded with the following instruments: IR: Nicolet–740 spectrophotometer, ¹H- and ¹³C-NMR: Varian Gemini 200 MHz and EIMS: VG Micromass 7070 H (70 eV). Optical rotations were determined with a Jasco DIP 360 Digital polarimeter.

The whole plants of *J. gossypifolia* were collected from the Osmania university campus in the month of May, 2000. The air dried plant material (2 kg) was extracted thrice with CH₂Cl₂–MeOH (1:1) (3 x 5 lit). Each extraction was continued for 3 days. The total extract was concentrated under reduced pressure to produce a thick brown gummy mass. The residue was subjected to column chromatography over silica gel (100–200 mesh) and the column was eluted

with solvents of increasing polarity using a mixture of hexane and EtOAc. The fractions eluted with hexane–EtOAc (2:3) were found (TLC) to contain the mixture of four compounds, which were separated by preparative TLC to afford jatrocin A (**1**, 12mg), jatrocin B (**2**, 15mg), propacin (**3**, 23 mg) and cleomiscosin A (**4**, 37 mg).

Jatrocin A (**1**): Viscous, $[\alpha]_{D25} \pm 0$ (c 1.02, MeOH); IR: ν_{\max} (KBr): 3417, 1710, 1612, 1584 cm^{-1} ; ^1H - and ^{13}C -NMR: Tables 1 and 2; EIMS: m/z 356 (M^+), 312, 194, 164, 137; Found: C, 64.17; H, 4.43. $\text{C}_{19}\text{H}_{16}\text{O}_7$ requires: C, 64.04; H, 4.49%.

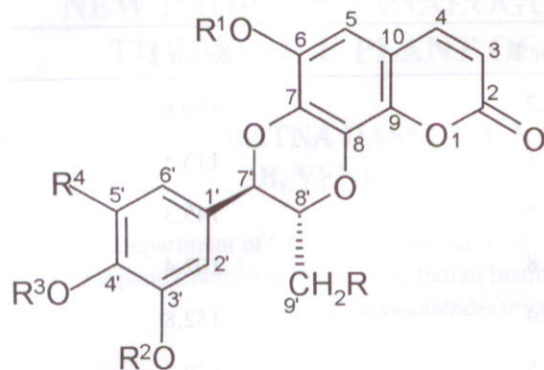
Jatrocin B (**2**): Viscous, $[\alpha]_{D25} \pm 0$ (c 1.31, MeOH), IR: ν_{\max} (KBr): 3406, 1710, 1618, 1595 cm^{-1} ; ^1H - and ^{13}C -NMR: Tables 1 and 2; EIMS: m/z 400 (M^+), 356, 208, 194, 167; Found: C, 62.85; H, 5.09. $\text{C}_{21}\text{H}_{20}\text{O}_8$ requires: C, 63.00; H, 5.00%.

Propacin (**3**): White solid, m.p. 227–228°C; $[\alpha]_{D25} \pm 0$ (c 1.12, MeOH). Spectral properties were similar to those reported^{11,12} earlier.

Cleomiscosin A (**4**): White solid, m.p. 244–245°C $[\alpha]_{D25} \pm 0$ (c 1.09, MeOH). Spectroscopic data were identical to the reported values.¹³

RESULTS AND DISCUSSION

Jatrocin A (**1**) was obtained as viscous mass and it was optically inactive. Its molecular formula was derived as $\text{C}_{19}\text{H}_{16}\text{O}_7$ from its elemental analysis and mass spectrum. The IR spectrum of the compound (vide Experimental) indicated the presence of hydroxy group, coumarin moiety and unsaturation in the molecule. The ^1H - and ^{13}C -NMR spectral data (Tables 1 and 2) were closely related to those of the known coumarino-lignan, propacin (**3**)^{11,12} isolated together with **1**. The structural difference between the compounds **1** and **3** was observed that the former contained two hydroxyl groups (δ 8.52 and 8.48, each brs, 1H each) and a methoxyl (δ 3.85, 3H, s) but the latter contained one hydroxyl and two methoxy groups. 2D-NMR spectral studies on jatrocin A (**1**) showed that the proton of one of hydroxy group (δ 8.48) was related to C-5 (δ 101.8) and C-7 (δ 138.4) while the proton of other hydroxy group (δ 8.52) was related to C-3' (δ 149.4) and C-5' (δ 115.6) in the HMBC experiment. Thus one of the hydroxy groups of **1** is located at C-4' (like propacin) and the other at C-6 (instead of a methoxy group present in propacin). The HMBC experiment also showed the correlations between H-7' (δ 4.62, d, J = 8.0 Hz) and C-7 (δ 138.4) and again between H-8' (δ 4.28–4.16, m) and C-8 (δ 137.8) in **1**. The coupling constant between H-7' and H-8' was found to be 8.0 Hz indicating *trans* orientation of these two protons. However, as the molecule was optically inactive it was racemic like other coumarino-lignans.¹⁴ In the mass spectrum the most prominent peak was observed at m/z 164 (100%) (as was observed in the mass spectrum of propacin) resulting from the retro-Diels Alder cleavage of the molecule. Considering all these spectral values, the structure of jatrocin A (**1**) was derived as 6-demethylpropacin.



1 : $R = R^1 = R^3 = R^4 = H, R^2 = Me$

2 : $R^1 = R^2 = Me, R = R^3 = H, R^4 = OMe$

3 : $R^1 = R^2 = Me, R = R^3 = R^4 = H$

4 : $R^1 = R^2 = Me, R^3 = R^4 = H, R = OH$

Table 1. 1H -NMR spectral data of 1 and 2 in $CDCl_3 + DMSO-d_6$

Proton	δ_H for 1	δ_H for 2
H-3	6.30 (d, $J = 9.5$ Hz)	6.33 (d, $J = 9.5$ Hz)
H-4	7.62 (d, $J = 9.5$ Hz)	7.57 (d, $J = 9.5$ Hz)
H-5	6.57 (s)	6.54 (s)
H-2', H-5', H-6'	7.02 – 6.78 (m)	6.81(s)*
H-7'	4.62 (d, $J = 8.0$ Hz)	4.61 (d, $J = 8.0$ Hz)
H-8'	4.28 – 4.16 (m)	4.31 – 4.17 (m)
-Me	1.23 (d, $J = 7.0$ Hz)	1.28 (d, $J = 7.0$ Hz)
-OH	8.52 (brs) 8.48 (brs) 5.77 (brs)	8.49 (brs)
-OMe	3.85 (s)	3.82 3.84 (x 2)

* for H-2' and H-6' (2H), H-5' is absent (suggested from HMBC spectrum)

Table 2. ^{13}C -NMR spectral data of **1** and **2** in $\text{CDCl}_3+\text{DMSO}-d_6$

Carbon	δ_{C} for 1	δ_{C} for 2
C-2	160.7	160.6
C-3	114.1	113.5
C-4	144.8	145.3
C-5	101.8	102.4
C-6	151.4	152.8
C-7	138.5	139.2
C-8	137.8	137.6
C-9	142.4	141.7
C-10	112.2	111.9
C-1'	127.5	126.3
C-2'	111.2	106.5
C-3'	149.4	150.2
C-4'	148.2	135.4
C-5'	115.6	150.2
C-6'	120.8	106.5
C-7'	81.1	80.7
C-8'	73.4	72.8
-Me	17.7	18.3
-OMe	55.8	56.1 55.9

The second new compound, jatrocin B (**2**) was also isolated as viscous mass and it was optically inactive. Its molecular formula was deduced to be $\text{C}_{21}\text{H}_{20}\text{O}_8$ from elemental analysis and mass spectrum. The IR (vide Experimental), ^1H - and ^{13}C - NMR spectral data (Tables 1 and 2) clearly suggested that the structure of jatrocin B was related to that of propacin (**3**) as well as of jatrocin A (**1**). The former (**2**) was found to contain three methoxyls (δ 3.82 and 3.84 (x 2) in the ^1H -NMR spectrum and δ 55.9 and 56.1 (x 2) in the ^{13}C -NMR spectrum) instead of two methoxyls present in **3**. The two protons, H-2' and H-6' appeared as singlet (δ 6.81) indicating

that the positions C-3', C-4' and C-5' were substituted, 2D-NMR studies with the assistance of HMBC experiment suggested that in **2**, the proton of the hydroxyl group (δ 8.49, brs) was related to C-3' (δ 150.2) and C-5' (δ 150.2), both of which contained methoxy groups. That means, the hydroxy group is present at C-4'. The ^1H - and ^{13}C -NMR spectral values of all the protons and carbons respectively of the coumarin moiety of **2** were identical to those of the corresponding protons and carbons of propacin (**3**). The HMBC experiment on **2** also showed that H-7' (δ 4.61, d, J = 8.0 Hz) was correlated to C-7 (δ 138.2) and again H-8' (δ 4.31-4.17, m) to C-8 (δ 137.6). The coupling constant of H-7' and H-8' of the compound **2** was also 8.0 Hz supporting their *trans* orientation. However, the molecule was optically inactive and hence racemic in nature. The base peak in the mass spectrum appeared at m/z 194 (100%) due to retro-Diels Alder cleavage of the molecule. All these spectral data clearly established the structure of jatrocinn B (**2**) as 5'-methoxypropacin.

The known coumarino-lignans, propacin (**3**)^{11, 12} and cleomiscosin A (**4**)¹³ were identified by comparison of their physical and spectral properties with those reported earlier.

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