

Isolation and Identification of Iridoid Glucosides from Pedicularis flava Pall. **Growing in Mongolia**

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Abstract

We have isolated and structurally elucidated by spectroscopic methods the monoterpenoid iridoid glucosides mussaenoside and euphroside from the aerial parts of the Mongolian medicinal plant Pedicularis flava Pall. Euphroside and mussaenoside have been isolated for the first time from this natural source.

Keywords: Pedicularis flava; Iridoid; Euphroside; Mussaenoside; Monoterpenoid glucoside

Introduction

Mongolian medicinal plants still offer a great potential for the discovery of new drugs because most of them have not been studied in detail [1,2]. Only a few investigations focused on *Pedicularis flava* Pall. [3]. The flowers and leaves of this plant have been used for the treatment of joint inflammation and poisoning. Moreover, Pedicularis flava Pall. has been applied to wound healing. We have isolated the two monoterpenoid iridoid glucosides 1 and 2 (FIG.1) for the first time from Pedicularis flava Pall. and assigned their structures.

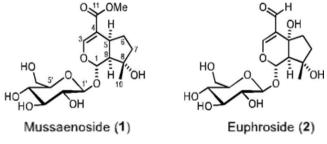
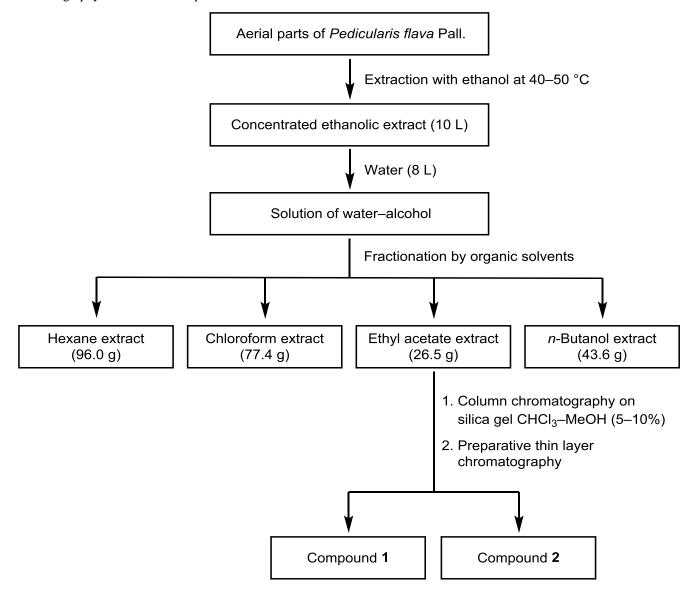


FIG. 1. Structures of the compounds 1 and 2.

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Isolation

The aerial parts of *Pedicularis flava* Pall. (Orobanchaceae) have been collected in Sergelen soum Tuv aimag, Mongolia, during the plant flowering period (August). The botanical assignment was made by Dr. Ch. Sanchir (Institute of Botany, Mongolian Academy of Sciences). The dried and grinded aerial parts of *Pedicularis flava* (12 kg) were extracted with ethanol (72 L) at 40°C to 50°C for 2 h (SCHEME 1). The alcoholic extract was filtered and concentrated using an evaporator at reduced pressure to give 10 L of a concentrated extract. Distilled water (8 L) was poured to the alcoholic extract to give a water–ethanol solution which was fractionated by extraction with solvents of increasing polarity: hexane, chloroform, ethyl acetate, n-butanol. The extracts were dried using Na₂SO₄. Evaporation of the solvents provided the hexane (96.0 g), chloroform (77.4 g), ethyl acetate (26.5 g) and butanolic extracts (43.6 g). Purification of the ethyl acetate extract by column chromatography on silica gel using chloroform–methanol (5% to 10%) as eluent and subsequent preparative thin layer chromatography afforded the compounds **1** and **2**.



SCHEME 1. Extraction and fractionation of Pedicularis flava Pall.

Results and Discussion

Compound 1 (Mussaenoside)

Compound 1 (mussaenoside) was obtained as an amorphous white powder. For the NMR spectra (500 MHz-¹H, 125 MHz-¹³C and DEPT, and HSQC), CD₃OD was used as solvent. On the basis of ESI-MS ($m/z = 391.4 [M+H]^+$ and 408.2 [M+NH₄]⁺) (FIG. 2) and ¹³C NMR spectra, compound 1 was assigned the molecular formula C₁₇H₂₆O₁₀. The ¹H and ¹³C NMR spectra confirmed that compound 1 is an iridoid monoterpenoid glucoside which by comparison with literature data [4–12] was found to be identical with mussaenoside (TABLE 1).

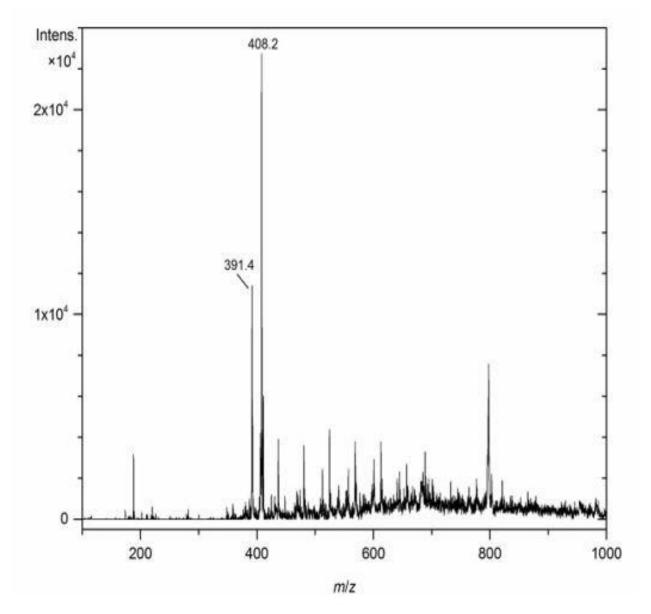


FIG. 2. ESI-MS (+10 V) of compound 1.

	Compound 1	Mussaenoside		Compound 1	Mussaenoside
	(500 MHz)	(400 MHz)		(125 MHz)	(100 MHz)
Н	δ (ppm), <i>J</i> (Hz)			δ (ppm)	
	Aglucone			Aglucone	
1	5.53 (1H, d, <i>J</i> = 4.2)	5.45 (1H, d, <i>J</i> = 4.2)	1	95.36 CH	95.4 CH
3	7.47 (1H, d, <i>J</i> = 1.0)	7.39 (1H, d, <i>J</i> = 1.0)	3	152.2 CH	152.1 CH
4	_	_	4	113.40 C	113.4 C
5	2.36 (1H, m)	2.28 (1H, m)	5	32.03 CH	32.1 CH
6	1.5 (2H, m)	1.43 (2H, m)	6	30.74 CH ₂	30.8 CH ₂
7	1.79 (2H, m)	1.7 (2H, m)	7	40.7 CH ₂	40.8 CH ₂
8	_	_	8	80.52 C	80.5 C
9	2.29 (1H, dd, <i>J</i> = 4.2, 9.2)	2.22 (1H, dd, <i>J</i> = 4.2, 9.2)	9	52.31 CH	52.4 CH
10	1.38 (3H, s)	1.32 (3H, s)	10	24.63 CH ₃	24.7 CH ₃
11	_	—	11	169.40 C=O	169.4 C=O
OMe	3.76 (3H, s)	3.69 (3H, s)	OMe	51.65 CH ₃	51.7 CH ₃
	Glucosyl			Glucosyl	
1′	4.74 (1H, d, <i>J</i> = 7.9)	4.67 (1H, d, <i>J</i> = 7.9)	1′	99.81 CH	99.9 CH
2′	3.27–3.45 (1H, m)	3.18 (1H, dd, <i>J</i> = 7.9, 9.2)	2'	74.74 CH	74.8 CH
3'	3.27–3.45 (1H, m)	3.24 (1H, dd, <i>J</i> = 8.9, 9.2)	3'	78.42 CH	78.4 CH
4′	3.27–3.45 (1H, m)	3.36 (1H, t, <i>J</i> = 9.2)	4'	71.72 CH	71.8 CH
5'	3.27–3.45 (1H, m)	3.18 (1H, m)	5'	78.01 CH	78.1 CH
6'	3.71 (1H, dd, <i>J</i> = 6.4, 12.0); 3.97 (1H, dd, <i>J</i> = 2.0, 12.0)	3.64 (1H, dd, <i>J</i> = 6.1,11.9); 3.90 (1H, dd, <i>J</i> = 2.2, 11.9)	6'	62.93 CH ₂	63.0 CH ₂

TABLE 1. ¹H and ¹³C NMR data (CD₃OD) of compound 1 and comparison with mussaenoside.

Compound 2 (Euphroside)

For the NMR spectra (500 MHz-¹H, 125 MHz-¹³C NMR and DEPT, COSY, and HSQC), CD₃OD was used as solvent. A comparison of the ¹H and ¹³C NMR data of compound **2** with those reported for euphroside in the literature [9–15] confirmed that both compounds are identical (TABLE 2).

	Compound 2 (500 MHz)	Euphroside (500 MHz)		Compound 2 (125 MHz)	Euphroside (125 MHz)	
Н	δ (ppm)	δ (ppm), J (Hz)		δ (ppm)		
Aglucone				Aglucone		
1	5.97 (1H, s)	5.85 (1H, br s)	1	95.34	95.3 CH	
3	7.41 (1H, s)	7.35 (1H, s)	3	163.09 CH	163.1 CH	
4	_	_	4	126.52 C	126.4 C	
5	_	_	5	71.24 C	71.3 C	
6	2.3 (1H, m); 2.15 (1H, m)	2.25 (1H, m); 1.90 (1H, m)	6	37.8 CH ₂	37.6 CH ₂	
7	2 (1H, m); 1.57 (1H, m)	2.1 (1H, m); 1.50 (1H, m)	7	40.33 CH ₂	40.3 CH ₂	
8		_	8	78.85 C	78.9 C	
9	2.5 (1H, s)	2.45 (br s)	9	61.46 CH	61.3 CH	
10	1.2 (3H, s)	1.2 (3H, s)	10	23.64 CH ₃	23.6 CH ₃	
11	9.3 (1H, s)	9.25 (1H, s)	11	192.53 CHO	192.6 CHO	
	Glucosyl			Glucosyl		
1'	4.69 (1H, d, J = 7.9)	4.62 (1H, d, J = 7.8)	1'	99.83 CH	99.8 CH	
2'	3.26 (1H, dd, J = 8.0, 9.3)	3.18 (1H, dd, J = 7.8, 9.0)	2'	74.36 CH	74.2 CH	
3'	3.43 (1H, m)	3.38 (1H, t, J = 8.5)	3'	78.29 CH	78.3 CH	
4'	3.34 (1H, m)	3.34 (1H, t, J = 8.5)	4'	71.65 CH	71.6 CH	
5'	3.31 (1H, m)	3.32 (1H, m)	5'	77.43 CH	77.3 CH	
6′	3.96 (1H, m); 3.73 (1H, m)	3.85 (1H, dd, J = 2.0, 11.7); 3.65 (1H, dd, J = 6.1, 11.7)	6′	62.81 CH ₂	62.7 CH ₂	

TABLE 2. ¹H and ¹³C NMR data (CD₃OD) of compound 2 and comparison with euphroside.

Experimental Section, General

NMR spectra were recorded on a Bruker DRX 500 spectrometer operating at 500 MHz for ¹H NMR and 125 MHz for ¹³C NMR. The chemical shifts δ are given in ppm; the solvent signals were used as reference (¹H: δ H = 3.35 ppm for residual CD₂HOD, ¹³C: δ c = 49.00 ppm). The coupling constants J are given in Hertz. ESI-MS were recorded on a Bruker-Esquire mass spectrometer with an ion trap detector. Positive and negative ions were detected.

Acknowledgements

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