

## Assessment of activated sludge respiratory inhibition of Bis (2-butoxyethyl) adipate

Karri.Apparao<sup>1</sup>, M.S.Surendra Babu<sup>2\*</sup> Tentu.Nageswara Rao<sup>1</sup>

<sup>1</sup>Department of Chemistry, Krishna University, Machilipatnam, AP, (INDIA)

<sup>2</sup>Department of Chemistry, GITAM University, Hyderabad, Telangana, (INDIA)

E-mail: tentu6581@rediffmail.com

### ABSTRACT

Effect of Bis (2-butoxyethyl) adipate was studied for its inhibitory effect on the respiration rate of an activated sludge under *in vitro* conditions. The method followed in the study was that described in the OECD guideline for testing of chemicals 209 (Section-4, adopted 4<sup>th</sup> April, 1984" Activated sludge, Respiration Inhibition Test"). The EC<sub>50</sub> (3h) of reference item 3,5-dichlorophenol was determined as 24.61 mg/L. The EC<sub>50</sub> (3h) of Bis (2-butoxyethyl) adipate was determined as 0.0841 g/L.

© 2015 Trade Science Inc. - INDIA

### KEYWORDS

Bis (2-butoxyethyl) adipate;  
3,5-dichlorophenol;  
Active sludge and  
inhibition.

### INTRODUCTION

The inhibitory effect on the respiration rate of microorganisms from an aerated biological clarification stage can be used as a measure of the toxicity of substances. The group of experts of ecotoxicology of the Organization for Economic Co-operation and Development (OECD) has published a method by which the inhibitory effect can be measured<sup>[1,2]</sup>. The "Effective concentration 50%" (EC<sub>50</sub>) is given as the value of measurement of inhibition effect<sup>[3,4]</sup>. The EC<sub>50</sub> value is the concentration of a substance or substance mixture at which inhibition of the respiration rate is 50%<sup>[5-8]</sup>. The method described assesses the effect of a test substance on micro-organisms by measuring the respiration rate under defined conditions in the presence of different concentrations of the test substance<sup>[9,10]</sup>. The purpose of this method is to provide a rapid screening method whereby substances which may adversely affect aerobic microbial treatment plants can be identified, and

to indicate suitable non inhibitory concentrations of test substances to be used in biodegradability tests<sup>[11]</sup>.

Bis (2-butoxyethyl)adipate is a high quality plasticizer for polar elastomers with an emphasis on low temperature applications, without compromising high temperature and volatility aspects. It is based on linear glycol ether chemistry. Compared to standard non-aromatic plasticisers, it has a better compatibility with medium to polar plastics, especially at low temperature and in contact with organic solvents and oils. Bis (2-butoxyethyl) adipate is part of a family of new plasticisers which can help in replacing phthalates in current applications. It can be used in nitrile, epichlorohydrine, polychloroprene or other fluorinated rubbers.

### EXPERIMENTAL

#### Materials and methods

Reference standard of Bis (2-butoxyethyl)adipate, 3,5 dichlorophenol were obtained from SigmaAldrich.

## Regular Paper

Peptone and Meat extract were purchased from the Himedia, Bombay. Urea, Sodium chloride, Calcium chloride, Magnesium sulphate, Dipotassium Hydrogen phosphate, Sodium hydroxide, Sulphuric acid AR Grade supplied by Merck Limited.

### Preparation of microbial inoculums

The sludge was collected from the Nijam Sugar Mill, Bobbili, Vizianagaram District, Andhrapradesh. The dry weight of the sludge was determined and it was enriched by adding synthetic waste-water (50 mL per litre of synthetic sewage sludge). The mixture was aerated overnight at  $20 \pm 2^\circ\text{C}$  and used for the study. For measurement of the sensitivity of sludge, the  $\text{EC}_{50}$  of the standard inhibitor substance 3,5-dichlorophenol was determined after a contact time of three hours.

## TEST METHODS

### General procedure

#### Cleaning and sterilization of glassware

All glassware were washed in cleaning solution and rinsed thoroughly in tap water. The cleaned glassware were sterilized by heating at about  $100^\circ\text{C}$  for 2 hours in hot air oven.

#### Measurement of temperature

The temperature of the test solution was recorded using calibrated thermometer.

#### Preparation of synthetic sewage feed

The following ingredients were dissolved in one liter of water.

Ingredients	Quantity (g)
Peptone	16.2
Meat extract	11.1
Urea	3.2
Sodium chloride (NaCl)	0.7
Calcium chloride ( $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ )	0.4
Magnesium sulphate ( $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ )	0.3
Dipotassium Hydrogen phosphate ( $\text{K}_2\text{HPO}_4$ )	2.9

#### Preparation of reference substance concentrations

Five hundred milligrams of 3,5-dichlorophenol was

dissolved in 10 mL of 1N NaOH and diluted to 30 mL with distilled water. Then 8 mL of 1N  $\text{H}_2\text{SO}_4$  was added by continuous stirring and finally the volume was made up to one liter with distilled water. The pH of the stock solution was 7.03.

Bottle No.	Treatments	Concentrations (mg/L)
1	Control 1	C1
2		5
3		10
4	3,5-dichlorophenol	20
5		40
6		80
7	Control 2	C2

#### Preparation of the test substance concentrations

The following concentrations of test item were weighed directly in reaction bottle for the analysis.

Bottle No.	Treatments	Concentrations (g/L)
8		0.1002
9		0.2502
10	Bis(2-butoxyethyl)adipate	0.5002
11		1.025
12		2.012
13	Control 3	C3

## TEST PROCEDURE

The following were added into 500 mL flat bottom flask and aerated for 3 h. 16 mL synthetic sewage feed X mL (5,10,20,40 and 80 mL) stock solution of 3, 5-dichlorophenol /x g (0.1002,0.2502, 0.5002, 1.025 and 2.012g) of Bis (2-butoxyethyl)adipate (water in case of controls) 200 mL aerated sewage sludge. Approximately 300 mL water and made up to 500 mL using water. After the end of 3h, the aeration was stopped. The decrease in oxygen was determined over a period of 10 minutes by reading off the oxygen concentration at every minute. The measurements of the oxygen concentrations were carried out with calibrated dissolved oxygen meter.

#### Calculation

The respiration rate was determined from the linear range of the oxygen decrease. From the rates of oxygen

decrease, the inhibition of the respiration rate of the sewage sludge was calculated by using the following formula:

$$\text{Percent inhibition} = \left\{ 1 - \left[ \frac{2R_s}{(R_{c1} + R_{c2})} \right] \times 100 \right\}$$

Where,

$R_s$  = Oxygen-consumption rate at tested concentration of test substance/ Reference item

$R_{c1}$  = Oxygen-consumption rate, Control 1

$R_{c2}$  = Oxygen-consumption rate, Control 2

**Validity criteria**

- 1 The respiration rates of two controls were within in the range (15%) of each other.
- 2  $EC_{50}$  value for the reference standard 3, 5

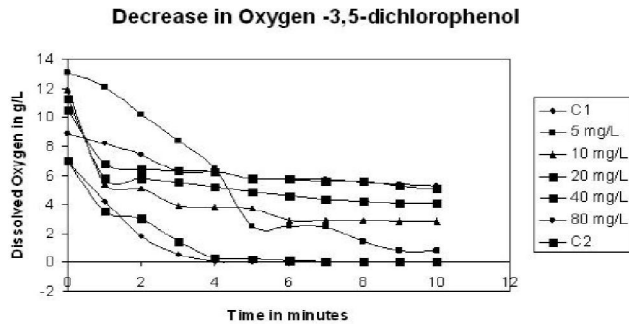


Figure 1 : Decrease in oxygen (3,5-dichlorophenol) over a period of ten minutes

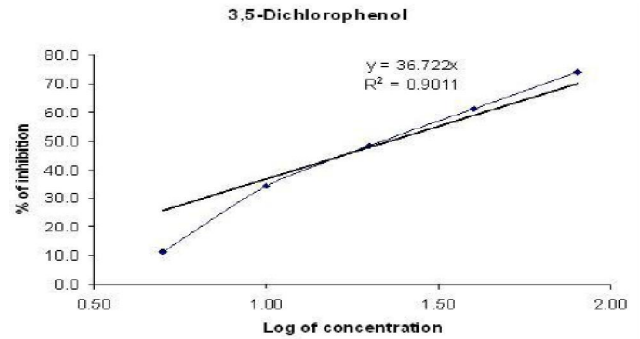


Figure 2 : Inhibition of the respiration rate (3,5-dichlorophenol)

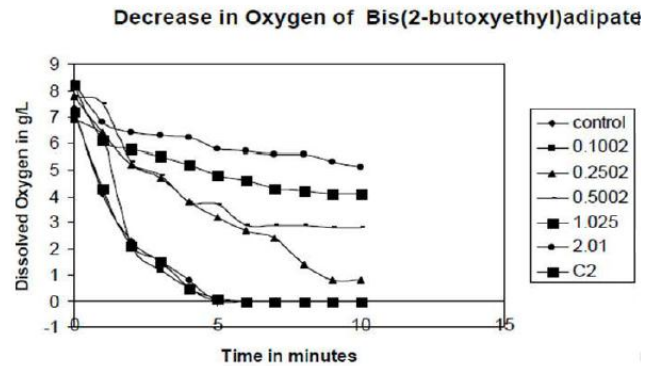


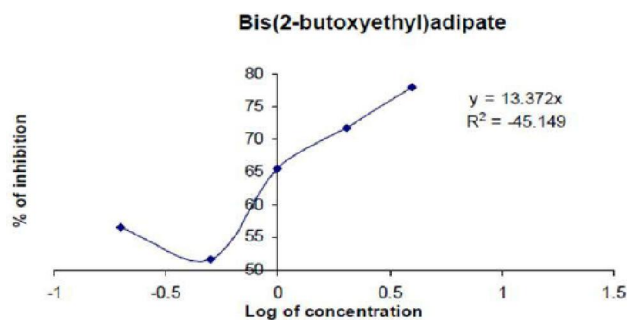
Figure 3 : Decrease in oxygen of bis (butoxyethyl) adipate over a period of ten minutes

- 3 The five test concentrations spaced by constant factors within the range 2 – 2.5 was used. This value assured the regulatory requirement.

TABLE 1 : Determination of The  $EC_{50}$  of 3, 5-Dichlorophenol and Bis(butoxyethyl) adipate

Reference Item						Test Item						
3,5-dichlorophenol (mg/l)						Proviplast 0142 (g/L)						
C1	5	10	20	40	80	C2	0.1002	0.2502	0.5002	1.025	2.012	C3
6.9	13.1	11.9	11.3	10.5	8.9	7	7.3	6.9	7.8	7.8	8.2	8.3
4.2	12.1	5.4	5.8	6.8	8.2	3.5	4.1	6.2	6.4	7.5	6.1	6.8
1.8	10.2	5.1	5.8	6.4	7.4	3	2.3	5	5.2	5.3	5.8	6.4
0.5	8.4	3.9	5.5	6.3	6.2	1.4	1.5	4.2	4.7	4.8	5.5	6.3
0	6.5	3.8	5.2	6.2	6.2	0.3	0.8	3.5	3.8	3.8	5.2	6.2
0	2.5	3.7	4.8	5.8	5.8	0.2	0	3.2	3.2	3.7	4.8	5.8
0	2.5	2.9	4.6	5.7	5.8	0.1	0	2.6	2.7	2.9	4.6	5.7
0	2.4	2.9	4.3	5.6	5.7	0	0	2.1	2.4	2.9	4.3	5.6
0	1.4	2.9	4.2	5.6	5.5	0	0	1.2	1.4	2.9	4.2	5.6
0	0.8	2.8	4.1	5.3	5.4	0	0	0.6	0.8	2.8	4.1	5.3
0	0.8	2.8	4.1	5.1	5.3	0	0	0.6	0.8	2.8	4.1	5.1
41.4	73.8	54.6	43.2	32.4	21.6	42	43.8	37.8	42	30	24.6	19.2
50.36	11.5	34.5	48.2	61.2	74.1	49.64	49.66	56.6	51.7	65.5	71.7	77.93
7.73	7.84	7.75	7.68	7.64	7.57	7.76	7.76	7.68	7.54	7.61	7.64	7.75

## Regular Paper



**Figure 4 : Inhibition of the respiration rate of bis (butoxyethyl) adipate**

### RESULTS AND DISCUSSION

The sludge contained 3.2 g/L (dry weight) suspended solids, filterable components. The decrease in oxygen concentration in the test preparation was linear over a wide range. Figure 1 shows the decrease in oxygen concentration after 3h exposure with reference standard 3, 5-dichlorophenol. The respiration rate was determined from the decrease in oxygen. Figure 2 shows the inhibition of the respiration rate against the concentration of 3,5-dichlorophenol. The  $EC_{50}$  of 3, 5 – dichlorophenol after an exposure time of three hours was determined as 26.6092 mg/L. Figure 3 shows the decrease in oxygen concentration after 3 h exposure time of Bis (2-butoxyethyl)adipate. Figure 4 shows the inhibition of the respiration rate against the concentration of Bis (2-butoxyethyl)adipate. The results are presented in TABLE 1. The  $EC_{50}$  after an exposure time of three hours was determined as 0.0841 g/L.

### CONCLUSION

From the results it was observed that the test item is inhibitory to the microbial organisms. The  $EC_{50}$  for

Bis (2-butoxyethyl)adipate after an exposure time of three hours was 0.0841 g/L.

### ACKNOWLEDGEMENT

The authors are thankful to the Dr. B. Gowtham Prasad, SVV University, for providing necessary facility to conduct the Laboratory experiment.

### REFERENCES

- [1] OECD, Activated Sludge; respiration Inhibition Test, Test Guideline No. 209, Guidelines for the testing of Chemicals., OECD, Paris, (1984).
- [2] ISO 8192, Water Quality- Test for inhibition of Oxygen consumption by activated sludge for carbonaceous and ammonium oxidation; International Organization for Standardization., (2007).
- [3] S.Fiebig, U.Noack; Fresenius Environmental Bulletin., **12b**, 1556-1557 (2004).
- [4] H.A.Painter; Toxicity Assessment., **1**, 515-524 (1986).
- [5] King EF, Painter HA; Toxicity Assessment., **1(1)**, 27-39 (1986).
- [6] Maazuza Z.Othman, Liqiang Ding, Yi Jiao; World Academy of Science., Engineering and Technology., **3**, 10-19 (2009).
- [7] M.Ahel; Wat.Res., **28(5)**, 1131-1142 (1994).
- [8] A.Di Corcia, R.Samperi; Environ Sci.Technol., **28**, 850-858 (1994).
- [9] D.Brown D, Hitz HR, Schafer L; Chemosphere., **10(3)**:245-261 (1981).
- [10] L.Luongo, X.Zhang; Journal of Hazardous Materials., **178**, 356-362 (2010).
- [11] Alex parise, Jackie Zhang; Journal of Nano Particle Research., **16**, 2159-2163 (2014).