



Susceptibility and curie temperature study in Cr^{3+} substituted Mg-Cd ferrites

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ABSTRACT

Temperature dependence normalized AC susceptibility and Curie temperature of spinel ferrites $\text{Cd}_x\text{Mg}_{1-x}\text{Fe}_{2-y}\text{Cr}_y\text{O}_4$ ($x=0, 0.2, 0.4, 0.8$ and 1.0 ; $y=0, 0.05$ and 0.1) study reveals that MgFe_2O_4 exhibits multi domain (MD) structure with high Curie temperature. On substitution of Cd^{2+} , MD to Single Domain (SD) transitions takes place and Curie temperature decreases due to decrease in A-B interaction. On substitution of Cr^{3+} in Mg-Cd ferrite system peak obtained in MgFe_2O_4 is suppressed which is attributed to decrease in grain size and further decrease in Curie temperature is attributed to the decrease in B-B interaction. This is because Fe-Fe interaction is greater than Cr-Fe interaction at B-site © 2014 Trade Science Inc. - INDIA

KEYWORDS

Normalized AC susceptibility;
Domain structure;
Curie temperature;
Magnesium ferrites.

INTRODUCTION

Susceptibility, a magnetic property of ferrites was found to depend upon grain size, grain boundary and domain structure^[1]. Ferrites consists of Multimomain (MD), single domain(SD) and superparamagnetic (SP) particles. The transition from one domain to another mainly depends on substitutions^[2]. These domain states can be studied by the technique of low field AC susceptibility^[3]. The MD particles contain domain walls (DW)^[4] and DW motion are responsible for magnetic changes. Formation of domain mainly depend on particle size. As particle size decreases, formation of domain walls becomes energetically unfavorable, then it is said to be single domain (SD) particle. In such a ferrites magnetic changes do not takes place through DW motion but require the rotation of spins resulting in larger coersivity. If particle size further decreases, spins are

affected by thermal fluctuations and the system becomes SP particle. SP particle nature reduces magnetic character of the material. Cd^{2+} substitution is interesting substitutioin in the spinels^[5]. Addition of Cr^{3+} in NiFe_2O_4 the domain structure changes from MD to SD^[6]. Al^{3+} substituted mixed Cu-Cd ferrites exhibit mixture of SD and MD partices^[7].

In the present investigation the efforts are made to understand the domain nature in Cr^{3+} substituted and unsubstituted Mg-Cd ferrite system using low field AC susceptibility measurements.

EXPERIMENTAL DETAILS

Polycrystalline spinel ferrite with general formula $\text{Cd}_x\text{Mg}_{1-x}\text{Fe}_{2-y}\text{Cr}_y\text{O}_4$ ($x=0, 0.2, 0.4, 0.6, 0.8$ and 1.0 ; $y=0, 0.05$ and 0.10) were prepared by standard ceramic method using AR grade oxides of Fe_2O_3 , MgO ,

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CdO and Cr₂O₃ (LOBA Chem. India). These pure oxides were accurately weighed accordingly to weight ratio required in the final proportions on single pan microbalance. The powders of each sample were mixed together and wet milled using acetone base. Dried powder of samples was prised at 700 °C for 12 hours and then sintered at 1050 °C for 24 hours. The pellets of samples were formed by applying 10⁶ Kg cm⁻² using hydrolic press. The pellets again sintered at 1050 °C for 24 hours for better compaction.

The powdered samples were characterised by XRD on philips computerised X-ray diffractometer (PW 3710) using Cuk α radiation. The AC susceptibility measurements of polycrystalline ferrite sample was made on Helmholtz double coil set up operated at 260 Hz with constant field of 70e, in the temperature range 300K to 800K. Platinum–Rhodium thermocouple was used to measure temperature of the powder sample. The Curie temperature of all the pelletized samples was measured by using modified Lorria-Sinha method.

RESULTS AND DISCUSSION

Characterization

A study of X-ray diffraction reveals that all the compositions under investigation were face centered cubic spinel structure. Typical X-ray diffractogram is presented in Figure 1. It is found that lattice constant increases with Cd²⁺ concentration. Such increase in the lattice constant due to addition of cadmium content was reported in the literature^[5]. This increase in lattice con-

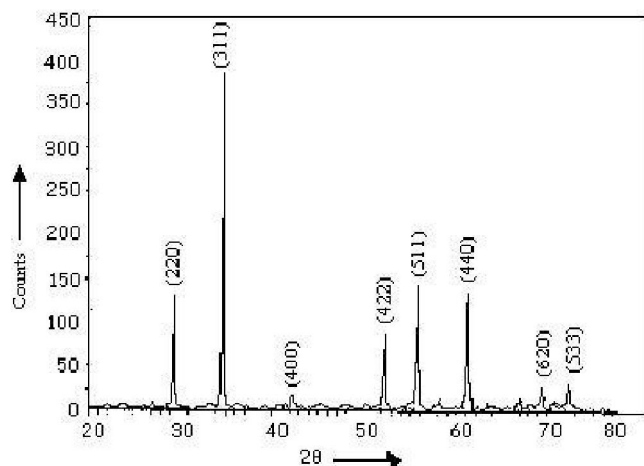


Figure 1 : Typical X-ray diffractogram of Cd_xMg_{1-x}Fe_{2-y}Cr_yO₄ ferrite with x = 0.6 y = 0.00.

stant is attributed to the difference in ionic radii of Cd²⁺ ion (1.03Å⁰) and Fe³⁺ion (0.67Å⁰). On substitution on Cr³⁺ ion, the lattice constant found to decrease. This was also attributed to the difference in ionic radii of Cr³⁺ ion (0.63Å⁰) and Fe³⁺(0.67Å⁰).

Normalized susceptibility

The plots of normalized susceptibility (χ/χ_{RT}) verses temperature are presented in the Figures.2-3. From these figures it can be seen that for magnesium ferrite, the susceptibility slowly increases and reaches peak value with temperature and suddenly drops to zero. The sudden drop of χ/χ_{RT} curve shows the formation of single phase cubic spinel^[8]. The increase in susceptibility with peak values suggests there is existence of multidomain particles in the material^[9]. The peak is found to suppressed with substitution of Cr³⁺ in MgFe₂O₄ and also Curie temperature (T_c) decreases with Cr³⁺ content. For the composition x = 0.2, y = 0, 0.05 and 0.1, susceptibility is found to be independent on temperature upto T_c and after T_c it suddenly drops to zero. Such nature of curve indicates that the presence of SD particles in the materials^[9]. Joshi et al^[10] also reported similar behaviour in Mg-Zn ferrite sys-

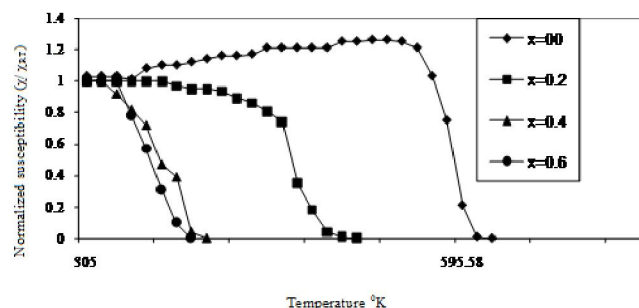


Figure 2 : The plots of normalized susceptibility (χ/χ_{RT}) verses temperature Cd_xMg_{1-x}Fe_{2-y}Cr_yO₄ Ferrite with y = 0.05.

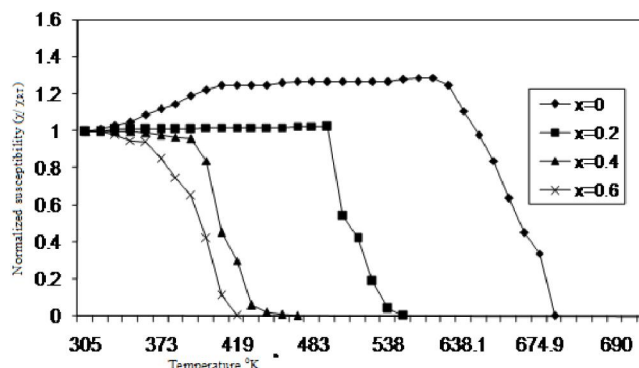


Figure 3 : The plots of Normalized susceptibility (χ/χ_{RT}) verses temperature Cd_xMg_{1-x}Fe_{2-y}Cr_yO₄ Ferrite with y = 0.10.

TABLE 1 : Curie temperature data from susceptibility measurement and Loria-Sinha method for the composition $Cd_xMg_{1-x}Fe_{2-y}Cr_yO_4$.

Conc.Cd ²⁺ (x)	Conc.Cr ³⁺ (y)	T _C from AC susceptibility measurement ^o K	T _C from Loria-SinhaMethod. ^o K
0.00		715	720
0.20		600	590
0.40	0.00	495	490
0.60		450	445
0.80		----	----
1.00		----	----
0.00		669	670
0.20		550	555
0.40	0.05	445	435
0.60		420	425
0.80		----	----
1.00		----	----
0.00		613	600
0.20		520	520
0.40	0.10	430	425
0.60		395	390
0.80		----	----
1.00		----	----

tem. The compositions with $x = 0.4$ and $x = 0.6$ for $y = 0, 0.05$ and 0.1 shows exponential decrease in susceptibility indicating SD to SP transition. The composition with $x = 0.8$ and $x = 1, y = 0, 0.05$ and 0.1 shows paramagnetic behaviour at and above room temperature. Curie temperatures (T_C) obtained from susceptibility plots are presented TABLE 1.

The Curie temperature measurement of all the samples was also been carried out by the method suggested by Loria-Sinha^[11] and also presented in the TABLE 1. These values are found to be in good agreement with the values obtained from temperature dependence of normalized susceptibility. On substitution of Cd²⁺ in MgFe₂O₄, Curie temperature was found to decrease. This is because substituted Cd²⁺ ion occupies tetrahedral (A) site, resulting into decrease in A-B interaction^[12]. The composition with $x = 0.8$ and 1.00 shows paramagnetic behaviour at room temperature, their Curie temperature lies below room temperature. Substitution of Cr³⁺ ion, Curie temperature of each composition is found to decrease. This is attributed to dilution of B-B interaction^[12]. On substitution Cr³⁺ ion occupies B-site replacing equivalent Fe³⁺ ions and so also decrease in magnetization at B-site.

CONCLUSIONS

Temperature dependent normalized susceptibility measurements reveals that MgFe₂O₄ exhibit MD particle and on substitution of Cd²⁺, domain structure changes from MD to SD and for higher concentration SD to SP. Curie temperature was found to decrease on substitution of Cd²⁺, which is attributed to the dilution of A-B interaction. On substitution of Cr³⁺, peak obtained in the graph of normalized susceptibility of MgFe₂O₄ is suppressed may be attributed to the decrease in grain size. Further decrease in Curie temperature in Mg-Cd ferrite system due to substitution of Cr³⁺ is attributed to the dilution of B-B site.

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