



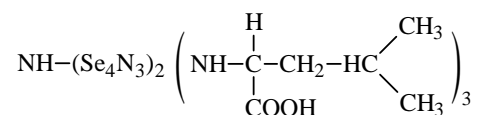
## MASS AND IR SPECTRAL STUDIES OF THE REACTION PRODUCT OF L-LEUCINE AND $\text{Se}_4\text{N}_3\text{Br}$

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### ABSTRACT

The reaction product of  $\text{Se}_4\text{N}_3\text{Br}$  with L-Leucine, an amino acid was synthesized in dichloroethane. The structure of the product is assigned on the basis of quantitative estimations, mol. wt., mass and IR spectrometric analysis as –



**Key words:**  $\text{Se}_4\text{N}_3\text{Br}$ , L-leucine.

### INTRODUCTION

The adducts of  $\text{Se}_4\text{N}_3\text{Cl}$  with phenylhydrazine, urea and thio-urea have been reported<sup>1,3</sup>, but the adducts of  $\text{Se}_4\text{N}_3\text{Br}$  with amino acids have not been synthesized till now. Therefore, it is intended to prepare the adducts of  $\text{Se}_4\text{N}_3\text{Br}$  with L-leucine and to investigate the reaction product, by mass and IR spectrometrically.

### EXPERIMENTAL

$\text{Se}_4\text{N}_3\text{Br}$  was prepared by the reaction of HBr on  $\text{Se}_4\text{N}_4$ , which was synthesized as reported<sup>1-7</sup> by the reaction of ammonia on  $\text{Se}_2\text{Cl}_2$  at 0°C. The adduct of  $\text{Se}_4\text{N}_3\text{Br}$  with L-Leucine was prepared by refluxing the equimolar mixture of both in dichloroethane for 6-8 h. The product, obtained, was separated, washed with dichloro-ethane, alcohol and ether, dried and stored in a vacuum desiccator over fused calcium chloride. The mass and IR spectra of the product were recorded on Jeol SX-102 (FAB) and Perkin Elmer RX1 (450-4000  $\text{cm}^{-1}$ ) spectrometers, respectively.

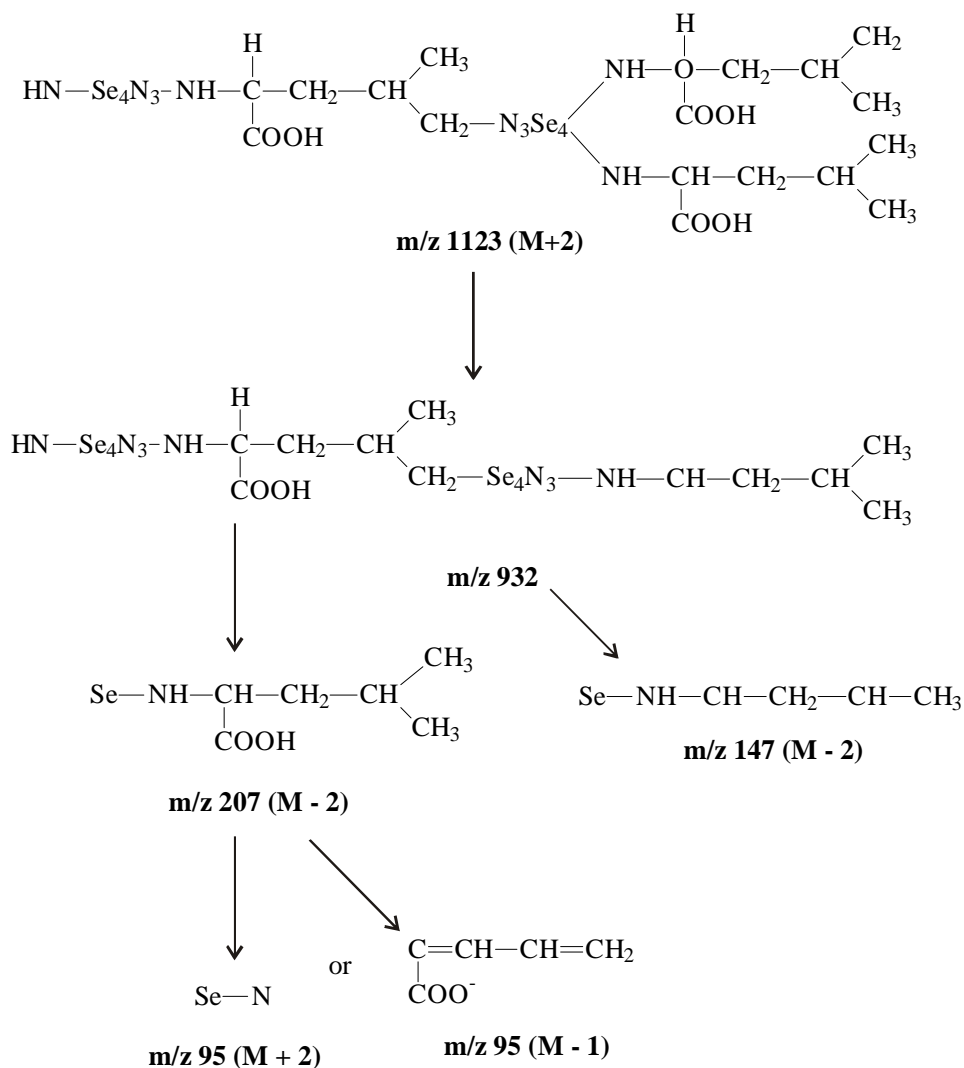
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## RESULTS AND DISCUSSION

The reaction product is yellowish orange solid, soluble in benzene. On the basis of quantitative estimation, % found (Cal.) Se 56.248 (56.348), N 12.456 (12.489), C 19.223 (19.269), O 8.543 (8.564), H 3.292 (3.301) and molecular weight 1120.6 (1121.0) g/mol. the adduct has been assigned as  $\text{HN}-(\text{Se}_4\text{N}_3)_2-(\text{NH}-\text{CH}-\text{COOH}-\text{CH}_2-\text{CH}(\text{CH}_3)_2)_3$ , which is supported by the prominent mass line at  $m/z$  1123 ( $M+2$ ) in its mass spectrum (Fig. 1).

The other mass lines in the mass pattern may be explained by FAB fragmentation process as follow:



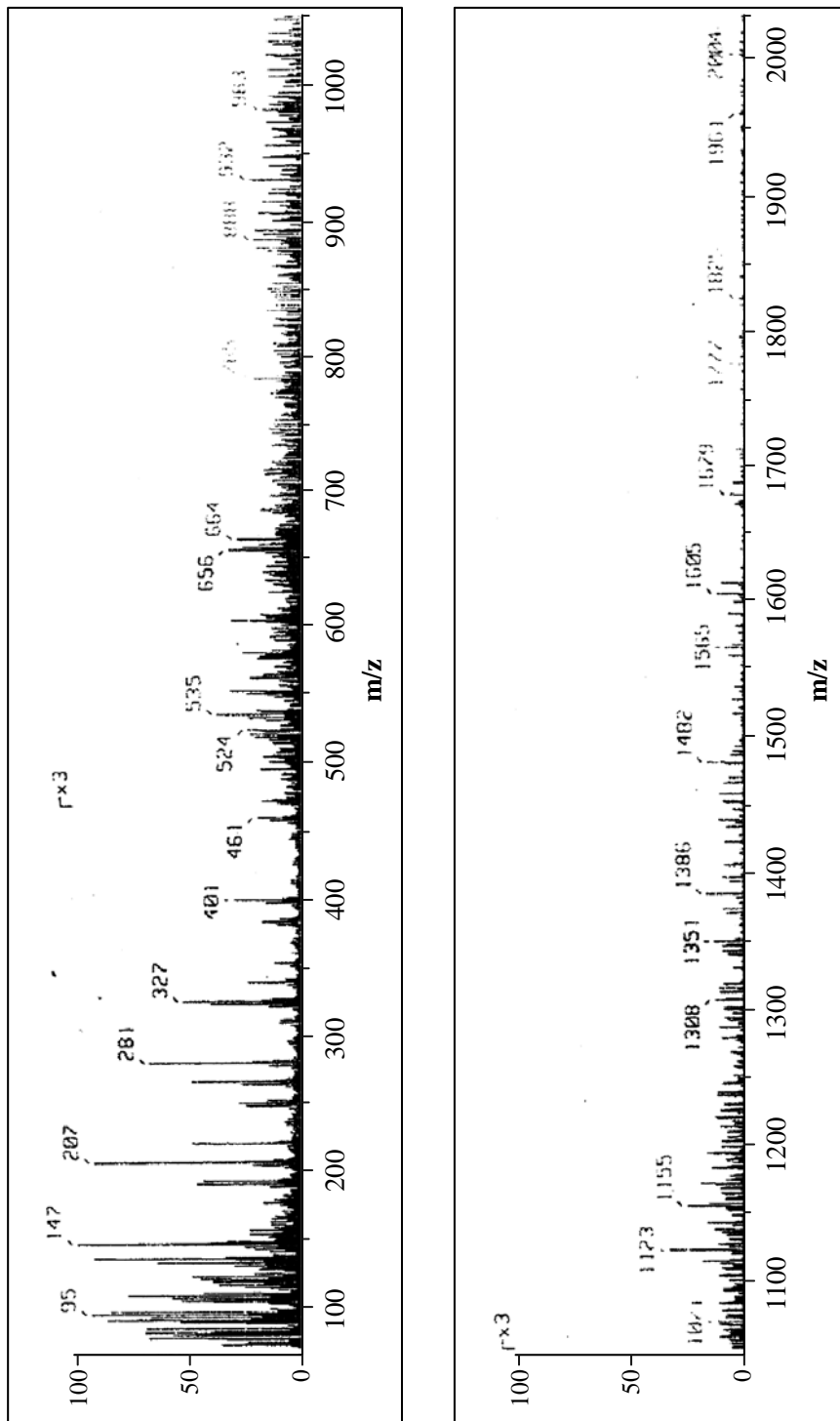
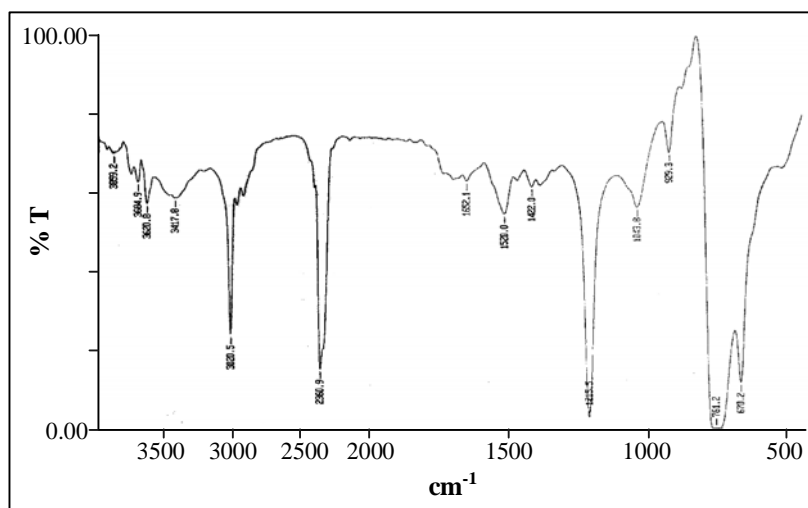


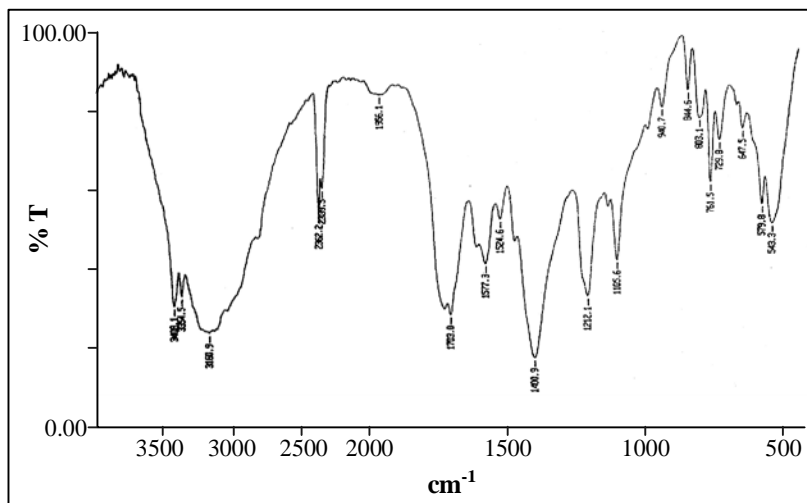
Fig. 1: Mass spectrum of Adduct



S. No.	Vibrations $\text{cm}^{-1}$		Bands assigned (c)	Force constant $\text{K} \times \text{N. (d)}$
	Ligand (a)	Product (b)		
7	1520.0	944.6 (s)	Se-N	6.2735
8	1652.1	940.7 (s)	Se-N	6.22182
9	2360.9	1105.6 (s)	C-O	4.955
10	3020.5	1212.1 (s)	$\text{CH}_3$	0.8012
11	3417.8	1400.9 (s)	$\text{CH}_3$	1.070
12	3620.8	1524.6	C-C	8.241
13	3684.9	1577.3 (b,d)	Se-N	17.49
14		1703.0 (b,d)	COOH	11.756
15		1956.0 (b,w)	Se-N	26.90
16		2339.5	Se=N	38.4823
17		2362.2	Se=N	39.2327
18		3160.9 (b)	N-H	5.449



**Fig. 2(a): IR Spectrum of Ligand ( $\text{Se}_4\text{N}_3\text{Br}$ )**

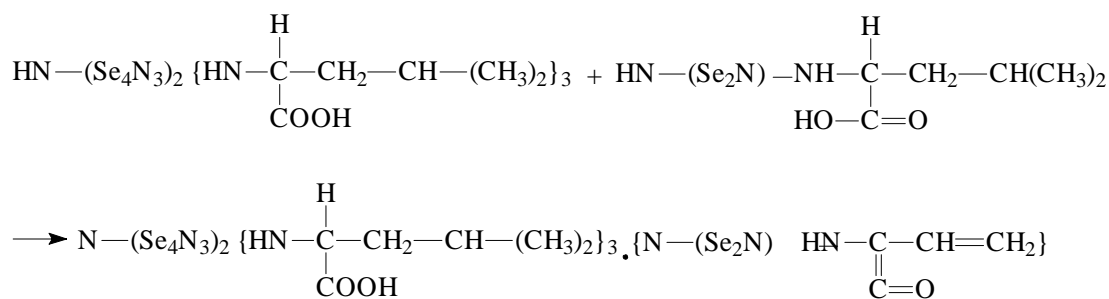


**Fig. 2(b): IR Spectrum of Adduct**

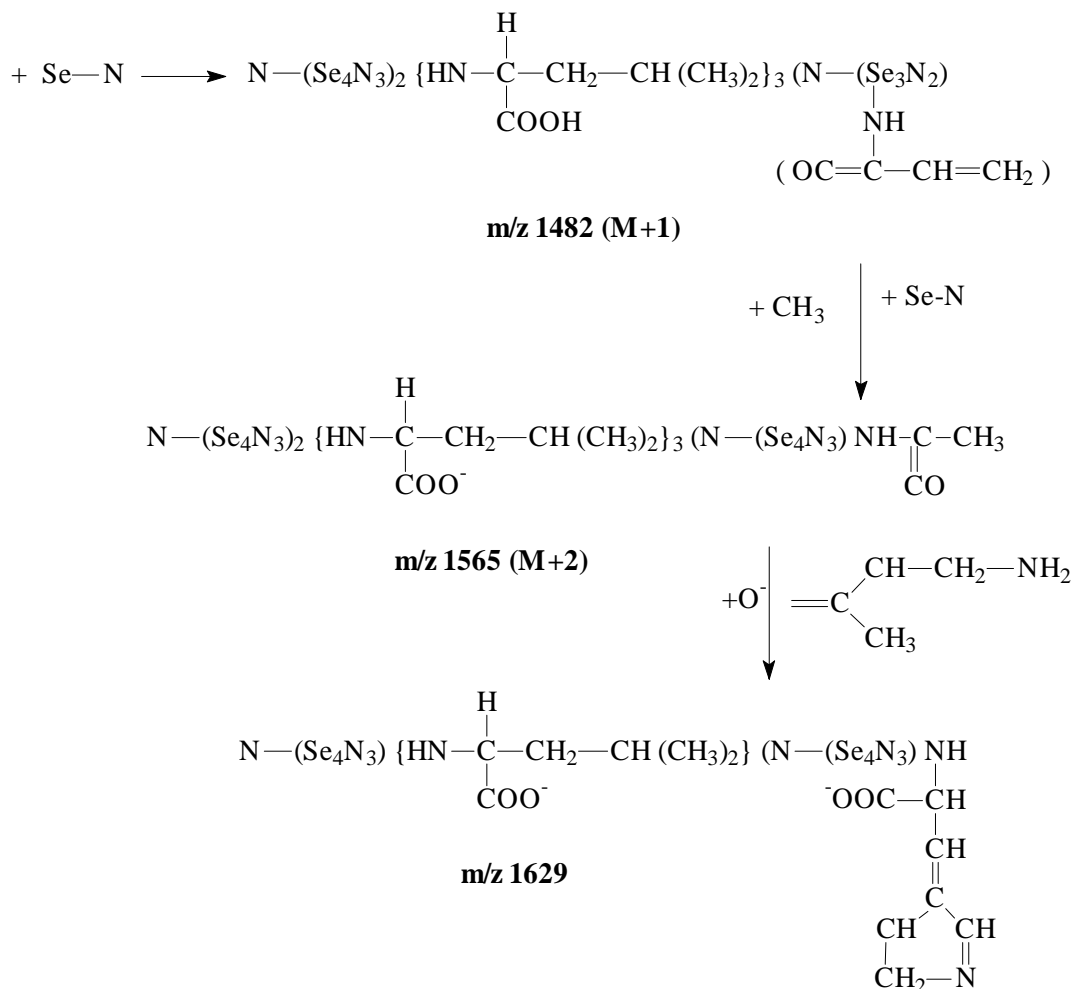
The appearance of other mass lines at  $m/z$  1386, 1482, 1507 and 1629 in its mass pattern may be impounded on the basis of recombination of Leucine phosphazanide and other fragments formed as follows:

The vibrations appeared in I.R. spectrum (Fig. 2, Table 1) at  $543.3\text{ cm}^{-1}$ ,  $579.8\text{ cm}^{-1}$ ,  $647.5\text{ cm}^{-1}$ ,  $729.8\text{ cm}^{-1}$ ,  $761.5\text{ cm}^{-1}$ ,  $803.1\text{ cm}^{-1}$ ,  $944.6\text{ cm}^{-1}$ ,  $940.7\text{ cm}^{-1}$  are for the Se-N bands while the vibration  $1105.6\text{ cm}^{-1}$  for C-O band and  $1212.1\text{ cm}^{-1}$ ,  $1400.9\text{ cm}^{-1}$  for  $\text{CH}_3$  group,  $1524.6\text{ cm}^{-1}$  for C-C band and  $1703.0\text{ cm}^{-1}$  for COOH group,  $3160.9\text{ cm}^{-1}$  for C-H band and  $3354.5\text{ cm}^{-1}$ ,  $3408.1\text{ cm}^{-1}$  for N-H band, of the L. Leucine amino acid, showing its presence and linkage in the adduct along with  $\text{Se}_4\text{N}_3^-$  ion.

The results revealed that  $\text{Se}_4\text{N}_3\text{Br}$  has reacted with L-Leucine an amino acid with formation of adduct and eliminating HBr during their reaction as mentioned Scheme 1.



**$m/z$  1386 (M - 2)**

**Scheme 1**

The presence of Se – N and Se = N bands in  $\text{Se}_4\text{N}_3$  ring is also inferred by the values of force constants (Table 1), calculated from the frequencies appear in its I.R. spectrum.

### ACKNOWLEDGMENT

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